

## Tier 2 Detailed Site Investigation Former Landfill at Tipperary Town



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## 1. INTRODUCTION

South Tipperary Council (the Council) completed a Tier 1 Assessment of the closed Tipperary Town Landfill in accordance with the "Code of Practice Environmental risk Assessment for Unregulated Waste Disposal Sites (CoP)" published by the Environmental Protection Agency (EPA).

The Assessment concluded that the site was a Class A – High Risk site, due the potential for leachate migration to surface water and the risk to humans from landfill gas linked to the nature of the bedrock beneath the site.

In September 2009, the EPA prepared guidance on the completion of the Tier 2 Assessment, which recommended that it be completed in two phases. Phase 1 should consist of Exploratory Site Investigation Works, following which the findings of the Tier 1 Assessment should be revised and the need for and/or extent of a Phase 2 Detailed Site Investigation should be determined in the second state of the second

The Department of Environment Heritage and Local Government (DEHLG) advised all local authorities that the Tier of Assessment would require an input from consultants with experience in the investigation of waste disposal sites, and in particular risk assessment. The Council appointed O'Callaghan Moran & Associates (OCM) to complete the Tier 2 Assessment.

OCM completed the Exploratory Site Investigation in November 2009. Given the high risk ranking, OCM considered that a geophysical survey should be completed, although this is not recommended for the Exploratory Phase in the EPA guidance. The survey provided valuable information on the site and in particular

- Allowed a more comprehensive delineation of the lateral and vertical extent of the waste;
- Confirmed the presence of a leachate plume migrating toward the surface water drain to the east of the site;
- Identified pockets of metal buried in the waste, which could present difficulties in subsequent drilling;
- Confirmed the composition of waste types and distribution and established total thickness of waste ;
- Allowed estimates of the thickness of subsoil beneath the waste and depth to bedrock;
- Indicated that the bedrock was most likely a shaley limestone, which was not a Regionally Important Aquifer.

The investigations confirmed the Class A High Risk Site category. However, the highest risk was linked to leachate migration to the surface water system. The risk presented by landfill gas to site users was eliminated, as it was proposed to demolish the onsite buildings. The level of risk posed to nearby residences reduced from High to Moderate, based on the information on the nature of the underlying bedrock obtained from the geophysical survey.

OCM concluded it was likely that leachate was migrating from the site toward a wetland area and into a drain, which ultimately discharges to the River Ara, several kilometres downstream of the site. OCM also concluded that a Detailed Site Investigation was required to assess the risk posed by leachate migration to the shallow groundwater system and the surface water system and the risk of off-site migration of landfill gas.

The Council submitted the OCM Exploratory Works report to the EPA for comment. The EPA accepted OCM's conclusions and recommended that the Detailed Site Investigation should include works to confirm the nature of the bedrock beneath the site.

#### 1.1 Work Scope

ses only any other use OCM scoped out the Detailed Site Investigation based on the results of the Exploratory Works and EPA's comments? A network of deep bedrock groundwater monitoring wells was not considered necessary, but one borehole should be installed to confirm the nature of the bedrock. Monitoring wells were required to monitor the shallow groundwater in subsoil zone, where potential leachate migration had been detected during the geophysical survey. The proposed works included:

- Review of surface water quality in the drain leaving the site (results not available for exploratory phase assessment).
- Installation of three groundwater monitoring wells in the subsoils.
- Installation of one borehole to the bedrock.
- Installation of three combined leachate/landfill gas monitoring wells in the waste body.
- Installation of five gas monitoring wells outside the waste body, three of which would also be suitable for groundwater monitoring.
- **Topographical Survey**

In response to EPA's comments, the findings of the geophysical survey were reevaluated to determine the validity of the conclusion on the nature of the bedrock aquifer.

### 2. SITE DESCRIPTION

The site is in the Townland of Carrownreddy and is within the Tipperary Environs area Figure 2.1). It served as the landfill for Tipperary Town from circa 1940, until it closed l in 1990. It is currently used by Tipperary Town Council as a Depot for road maintenance materials and machinery.

The site is approximately 1.8 hectares and contains a fenced off area of 0.2 hectares, which was apparently used exclusively for the disposal of wastewater sludge. The southern, and part of the eastern and western boundary is fenced, but there is no visible boundary, other than the raised landfilled area, on the northern side. In addition to the sludges, the other wastes accepted were most likely to have been commercial and domestic.

The southern part of the site has a hardcore surface and is used for storing road maintenance materials and machinery. The remainder of the site, north of the shed is covered with miscellaneous wastes, including farge mounds of construction & demolition waste, waste tyres, household waste, white goods (fridges, washing machines etc) and green waste.

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#### 2.1 Surrounding Land Use

The adjoining lands are currently used primarily for low intensity agriculture, (grazing horses). Immediately to the north is a marshy area. There are at least 20 residences within 250m of the north west and western site boundaries. There is also a newly developed housing estate located approximately 250m to the south eats of the site. A residential development (~250 houses) is under construction approximately 200m to the northeast of the site and it is the intention to develop the land between landfill and the residential estate for light industrial warehousing.

The lands to the south are currently used for grazing, but it is intended to develop these lands for social housing and light industrial use. There are currently no proposals to develop the lands to the west, but there are plans to extend the Lake Road west to link up with the R497, the Donohill Road.

Tipperary Town Council intends to move the Depot to an alternative location to enable the investigation and remediation of the site. In the longer term, the Council intends to leave the site as a closed landfill.

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The intrusive site investigation works, which included the installation of the groundwater water and landfill gas monitoring wells and the collection of surface water and groundwater samples for laboratory testing, were carried out between 16<sup>th</sup> and 20<sup>th</sup> November 2009.

The investigations were undertaken in accordance with BS10175, 2001, Investigation of Potentially Contaminated Sites - Code of Practice and were supervised by OCM personnel experienced in the investigation of landfills and contaminated lands. Ms Ruth Hennessy of the Council attended on site during the works.

**3.1 Well Installation** OCM provide specifications for the monitoring wells to the drilling contractor, who was experienced in the investigation of waste disposal sites. A track mounted air rotary drill rig, capable of travelling on variable terrain, was used to install the wells. The boreholes were logged in accordance with BS5930. The borehole locations are shown on Figure 3.1 and borehole logs are included in Appendix 1.

#### Leachate Landfill Gas Wells 3.1.1

Three combined leachate/landfill gas wells (MW-1, MW-2 and MW-3) were installed in the waste body. The boreholes were advanced to the base of the waste and drilling stopped once the underlying natural ground was encountered. The depth to the base of the waste ranged from 10.4m to 13.5m below ground level (bgl). Leachate was encountered in the boreholes at depths ranging from 5.85m to 6.4m bgl.

#### 3.1.2 Groundwater/Landfill Gas Wells

Five groundwater/landfill gas wells (MW-4, MW-5, MW-6, MW-7 and MW-8) were installed outside the waste body to monitor groundwater quality and landfill gas. MW- 4, 5, 6, 7 and 8 were installed down hydraulic gradient of the landfill. MW -4 and MW-8 are to the east of the waste, MW-5 and MW-7 are to the south, MW-6 to the south west

The locations of MW-4 and 8 were based on the evidence of leachate migration identified in the geophysical survey. In particular, MW-4 was installed to see if leachate was reaching the surface water drain to the east of the site.

It was not possible to install boreholes in the lands to the north, northwest and northeast of the site because the marsh area was flooded and the soft ground conditions prevented safe access.

MW-4, 5, 6 and 8 were advanced to an average depth of 10m below ground level. MW-7 was advanced to a depth of 20m below ground level. This borehole was installed to try to prove the depth and nature of the bedrock.

The drilling proved 0.65m of soft brown clay underlain by up to eight metres of very stiff brown clay with occasional boulders, which are dry. In MW-7 the clay is underlain by a layer of clay with gravel that extends to a depth of 13.5m and are water bearing. These clayey gravels are underlain by sand and gravel to 14.1m. A layer of dry very stiff clay underlies this from 14.1m to 15.75m. Beneath this is a water bearing layer of gravels from 15.75 to 20m (Ref Table 3.1).

It was not possible to drill beyond 20m because the gravels prevented the casing from advancing by jamming the drill stem against the casing.

To prevent MW-7 from becoming a conduit for vertical migration of water from the upper to the fower gravel zone the borehole was back filled with a concrete/bentonite sturry using a tremie pipe to plug the borehole from the base to a level above the clay layer. A summary of the subsoil profile is presented in Table 3.1.

Table 3.1Subsoil Profile Summary

Depth (m)	Description
0-0.65	Soft brown Clay.
0.65 – 7.35	Brown very stiff sandy gravelly CLAY. Gravels are subangular dark grey limestone.
7.35 - 13.35	CLAY and GRAVEL. Gravels are subangular to subrounded. Groundwater strike at
	8.85m.
13.35 - 14.1	Dark grey limestone SAND and GRAVEL.
14.1 - 15.75	Brown very stiff sandy gravelly CLAY. Gravels are subangular dark grey limestone.
15.75 - 20	Clay rich dark grey limestone GRAVELS. Gravels are subangular to subrounded.
	Groundwater strike at 15.75m.

\* Subsoil profile based on the borehole log for MW-7

#### 3.1.2.1 Well Construction

The leachate and landfill gas monitoring wells were constructed using high density polyethylene (HDPE) 50 mm diameter standpipes, which were slotted from the base of the hole (the base of the waste material) to approximately 1m below ground level.

A gravel filter pack was inserted in the annular space between the boring and the standpipe to a level of 0.5 m above the slotted section of the standpipe. Above the gravel filter the annular space was filled with a bentonite seal. The solid section of the well pipe was brought above the ground level and was fitted with a landfill gas cap and valve to allow landfill gas monitoring. A steel protective well casing, set in a concrete base, was placed around each standpipe.

The groundwater wells were constructed using HDPE 50 mm diameter standpipes which were slotted from the base of the hole to approximately 2m below ground level.

A gravel filter pack was inserted in the annular space between the boring and the standpipe to a level of 0.5 m above the slotted section of the standpipe. Above the gravel filter the annular space was filled with a bentonite seal. The solid section of the well pipe was brought above the ground level and was fitted with a landfill gas cap and valve to allow landfill gas monitoring. A steel protective well casing, set in a concrete base, was placed around each standpipe.

#### 3.2 Groundwater Monitoring

#### 3.2.1 Sampling Methodology

Groundwater samples were taken from wells MW-4 to MW-8 on the 23<sup>rd</sup> November 2009. The samples were collected in accordance with the OCM groundwater sampling protocol, which is included in Appendix 2.



	O' Callaghan Moran & Associates. Granary House, Rutland Street, Cork, Ireland. Tel. (021) 321521 Fax. (021) 321522	CLIENT SOUTH TIPPERARY COUNTY COUNCIL	DETAILS	Figure No. 3.1
environmental management for business	email : info@ocallaghanmoran.com	TITLE		SCALE
This drawing is the property of not be used, reproduced or dis permission of O'Callaghan Moran &	O'Callaghan Moran & Associates and shall sclosed to anyone without the prior written Associates and shall be returned upon request.	TRIAL PIT AND BOREHOLE LOCATIONS		1:1,000

After completion of groundwater level measurements, each well was purged to remove the stagnant water in the well and surrounding gravel pack using a 12 volt submersible pump and dedicated polypropylene tubing in each well to prevent cross contamination. pH, electrical conductivity and temperature were measured and the results along with visual observations are presented in Table 3.2.

The samples were placed in laboratory prepared containers, stored in a cooler, and sent for analyses to Jones Environmental Forensics Ltd laboratory. The Chain of Custody (COC) documentation is included in Appendix 3.

Borehole Number	MW-4	MW-5	MW-6	MW-7	<b>MW-8</b>
Water Level (mBTOC)	0.3	1.85	0.7	6.35	3.94
Top of Casing (mOD)	93.26	96.71	94	<sup>3</sup> 95.59	93.2
Water Level (mOD)	92.96	94.86	993.314 OL	89.24	89.26
рН	7.98	8.22 purp	vine8.16	7.99	7.65
EC (µS/cm)	1,365	1, 120 The Met	1,320	1,102	1,398
Temperature	10.1	5100910.2	10.2	10.3	10.6
Colour	Cloudy	Cloudy	Cloudy	Clear	Clear
Odour	None	None	None	None	None
Recovery	Good	Good	Good	Good	Good

**Table 3.2**Groundwater Field Measurements

#### 3.2.2 Laboratory Analysis

The samples were analysed for a range of organic and inorganic parameters that included, pH and electrical conductivity, dissolved oxygen, ammonia, nitrite, nitrate, orthophosphate, potassium, sodium, chloride, sulphate, metals, cyanide, total organic carbon (TOC), polyaromatic hydrocarbons (PAH), volatile organic compounds (VOC), semi-volatile organic compounds (sVOC) and pesticides.

The laboratory methodologies were all ISO/CEN approved or equivalent and the method detection limits (MDL) were all below the relevant guidance limit.

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#### 3.2.3 Laboratory Results

The full laboratory test report is in Appendix 3 and the results are summarised in Table 3.3, 3.4 and 3.5. The table includes Interim Guideline Values (IGV) published by the EPA. The IGVs are not statutory, but were developed to assist in the assessment of impacts on groundwater quality in the context of the implementation of the EU Water Framework Directive. The guidelines are based on, but are more conservative than the Dinking Water quality standards.

Sample I.D.	Unite	MW-4	MW-6	MW-8	ICV	
Sample Date	Omts	101 00 -4	101 00 -0	141 44 -0	10,	
Arsenic	μg/l	6	6.6	6.6	10	
Boron	μg/l	25	258	20	1,000	
Cadmium	μg/l	< 0.5	<0.5	< 0.5	5	
Copper	μg/l	<7	<7	12	30	
Mercury	μg/l	<1	<1	<u></u> <1	1	
Nickel	μg/l	<2	2 2	<sup>151</sup> 4	20	
Lead	μg/l	5	att. Tay ou	8	10	
Zinc	μg/l	<3 🥰	ator<3	10	100	
Iron	μg/l	<20 %	x <sup>c</sup> <20	<20	200	
Manganese	μg/l	ectil 18	342	538	50	
Calcium	mg/l	19.10	144.4	147.7	200	
Magnesium	mg/lov	9.30	14.82	19.03	50	
Sulphate	mg/l	14.78	104.22	11.22	200	
Chloride	Constraints/l	57.9	135.9	276.2	30	
Fluoride	mg/l	< 0.3	< 0.3	< 0.3	1	
Total Alkalinity as	mg/l	308	388	368	No Abnormal	
Total Cyanide	111g/1	-40	-40	-40		
Chromium - total	μg/1	<1.5	~1.5	<15	30	
Phosphorous	μg/1	10	12	11	30	
Potassium	$\frac{\mu g}{l}$	1 58	5.64	1 21	5	
Sodium	mg/l	40.11	101 30	81.15	150	
рН	pH units	7.82	8.02	8.30	6.5-9.5	
Electrical Conductivity	μS/cm	1232	1389	1490	1.000	
Total Oxidised Nitrogen	mg/l	1.22	42.27	< 0.05	No Abnormal Change	
Ammonia	mg/l	0.7	0.8	0.2	0.15	
Total Dissolved Solids	mg/l	472	947	919	-	
TOC	mg/l	5	10	6	-	

 Table 3.3 Groundwater Results – Inorganics and Total Organic Carbon

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December 2009 (BS/SM)

Parameter	Units	MW-4	MW-6	<b>MW-8</b>	IGV
Naphthalene	μg/l	<0.1	<0.1	<0.1	1
Acenaphthylene	μg/l	< 0.08	<0.08	< 0.08	-
Acenaphthene	μg/l	< 0.1	<0.1	<0.1	-
Fluorene	μg/l	< 0.07	< 0.07	< 0.07	-
Phenanthrene	μg/l	< 0.07	< 0.07	< 0.07	-
Anthracene	μg/l	< 0.08	<0.08	<0.08	10000
Fluoranthene	μg/l	< 0.09	<0.09	< 0.09	1
Pyrene	μg/l	<0.12	<0.12	< 0.12	-
Benz(a)anthracene	μg/l	< 0.09	<0.09	< 0.09	-
Chrysene	μg/l	< 0.1	<0.1	<0.1	-
Benzo(bk)fluoranthene	μg/l	<0.26	<0.26	< 0.26	0.50
Benzo(a)pyrene	μg/l	<0.12	<0.12	< 0.12	0.01
Indeno(123cd)pyrene	μg/l	< 0.1	<0.1	<0.1	0.05
Dibenzo(ah)anthracene	μg/l	< 0.1	<0.1	<0.1	-
Benzo(ghi)perylene	μg/l	<0.12	<0.12	<0.12	-
2-Methylnaphthalene	μg/l	<10	<10	<10	-
<b>Total PAHS</b>	μg/l	<1.60 00	¥1.60	<1.60	

**Table 3.4 Groundwater Results PAH** 



## Table 3.5 Groundwater Results VOC, sVOC and Pesticides -

Parameter	Units	<b>MW-4</b>	<b>MW-6</b>	<b>MW-8</b>	IGV			
est <sup>io</sup> VOCs								
Ethylbenzene 🕻	<sup>on</sup> μg/l	<3	<3	<3	10			
p/m-Xylene	μg/l	<5	<5	<5	10			
o-Xylene	μg/l	<3	<3	<3	10			
1,2,4-Trimethylbenzene	μg/l	<3	<3	<3	-			
4-Isopropyltoluene	μg/l	<3	<3	<3	-			
Naphthalene	μg/l	<2	<2	<2	1			
DRO	mg/l	< 0.01	< 0.01	< 0.01	0.01			
Mineral Oil	mg/l	< 0.01	< 0.01	< 0.01	0.01			
sVOC	μg/l	ND	ND	ND	_			
Pesticides	μg/l	ND	ND	ND				

Elevated levels of ammonia manganese, chloride and electrical conductivity, which are indicative of leachate contamination were detected in the shallow gravel/clay zone The levels decrease in concentration moving from MW-8 located east of the waste body to MW-4, approximately 150m east of the landfill approximately 10 west of the drain.

#### 3.3 Leachate Monitoring

#### 3.3.1 Sampling Methodology

Leachate samples were collected from MW-1 to MW-3 on the 23<sup>rd</sup> November 2009. The samples were collected in accordance with the OCM sampling protocol, which is included in Appendix 2.

After completion of leachate level measurements, each well was purged using dedicated PVC bailers. The field measurements recorded are presented in Table 3.6. A strong hydrocarbon odour was noted from the sample collected from MW-1 and a black oily residue noted on the bailer. The samples were placed in laboratory prepared containers, stored in a cooler, and sent for analyses to Jones Environmental Forensics. The COC documentation is included in Appendix 3.

Borehole Number	MW-1	MW-1 MW-2		
Water Level (mBTOC)	Water Level 4.34		6.91	
Top of Casing 97.29 ion (mOD) sector		strouinee 8 ventime 98.47	98.59	
Water Level (mOD)	92.95 B	92.67	91.68	
рН	× 8.26	8.69	8.01	
EC	Const >3999	>3999	>3999	
Temperature	10.9	10.8	10.8	
Colour	Black	Black	Black	
Odour Hydrocarb		Putrescible	Putrescible	
Recovery	Good	Good	Good	

 Table 3.6
 Leachate Field Measurements

#### 3.3.2 Laboratory Analysis

The samples were analysed for a range of organic and inorganic parameters that included pH and electrical conductivity, total oxidised nitrogen (TON), ammonia, biological oxygen demand (BOD), chemical oxygen demand (COD), sulphate, chloride, fluoride, total alkalinity, metals, total cyanide, phosphorus, mineral oil, VOC, sVOC, PAH, diesel range organics (DRO) and pesticides. The laboratory methodologies were all ISO/CEN approved or equivalent and the method detection limits (MDL) were all below the relevant guidance limit.

#### 3.3.3 Laboratory Results

The full laboratory test report is in Appendix 3 and the results are summarised in Tables 3.7, 3.8 and 3.9. Included in the Table for comparative purposes are the groundwater IGVs

Sample I.D.	Units	MW-1	MW-2	MW-3	IGV
Sample Date	emus				
Arsenic	μg/l	19.2	17.1	10.3	10
Boron	μg/l	945	1917	733	1,000
Cadmium	μg/l	< 0.5	<0.5	< 0.5	5
Copper	μg/l	<7	<7	<7	30
Mercury	μg/l	<1	<1	<1	1
Nickel	μg/l	4	15	<2	20
Lead	μg/l	16	5	11	10
Zinc	μg/l	4	11	v <sup>se.</sup> 4	100
Iron	μg/l	81	52 othe	<20	200
Manganese	μg/l	903	s 01:985	706	50
Calcium	mg/l	122.30	47.91	166.40	200
Magnesium	mg/l	42,28	28.96	58.08	50
Sulphate	mg/l <sub>çð</sub>	115 dt 6.79	100.53	3.15	200
Chloride	mg/l&	235.2	948.6	1703.7	30
Fluoride	næg/l	< 0.3	0.3	0.5	1
Total Cyanide	μg/l	<40	<40	<40	10
Chromium - total	μg/l	19.1	2.5	16.6	30
Phosphorous	μg/l	56	336	21	30
Potassium	mg/l	74.02	127.00	65.60	5
Sodium	mg/l	100.30	352.50	586.30	150
рН	pH units	8.01	8.50	7.88	6.5-9.5
Electrical Conductivity	μS/cm	3710	4370	6370	1,000
TON	mg/l	< 0.05	< 0.05	< 0.05	No Abnormal Change
Ammonia	mg/l	70.5	43.5	18.1	0.15
BOD settled	mg/l	20	26	9	-
COD	mg/l	114	183	52	_

Table 3.7 Leachate Results – Inorganics, TON and BOD

Parameter	Units	MW-1	MW-2	MW-3	IGV
Naphthalene	μg/l	42.5	<0.1	<0.1	1
Acenaphthylene	μg/l	< 0.08	< 0.08	< 0.08	-
Acenaphthene	μg/l	1.4	<0.1	<0.1	-
Fluorene	μg/l	0.90	< 0.07	< 0.07	-
Phenanthrene	μg/l	0.80	< 0.07	< 0.07	-
Anthracene	μg/l	< 0.08	< 0.08	< 0.08	10000
Fluoranthene	μg/l	< 0.09	< 0.09	< 0.09	1
Pyrene	μg/l	< 0.12	< 0.12	< 0.12	-
Benz(a)anthracene	μg/l	<0.09	<0.09	< 0.09	-
Chrysene	μg/l	<0.1	<0.1	<0.1	-
Benzo(bk)fluoranthene	μg/l	<0.26	<0.26	< 0.26	0.50
Benzo(a)pyrene	μg/l	< 0.12	< 0.12	< 0.12	0.01
Indeno(123cd)pyrene	μg/l	<0.1	<0.1	<0.1	0.05
Dibenzo(ah)anthracene	μg/l	<0.1	<0.1	<0.1	-
Benzo(ghi)perylene	μg/l	<0.12	<0.12	<0.12	-
2-Methylnaphthalene	μg/l	19	<10 5	<sup>کور</sup> <10	-
<b>Total PAHS</b>	μg/l	64.60	<1.60 <sup>0</sup>	<1.60	

 Table 3.8 Leachate PAH Results - 23/11/2009



#### Table 3.9 Leachate VOC, sVOC and Pesticides Results

Parameter	Units	MW-1	<b>MW-2</b>	MW-3	IGV			
₹. VOCs								
Ethylbenzene	kg/l	4	<3	<3	10			
p/m-Xylene	μg/l	9	<5	<5	10			
o-Xylene	μg/l	5	<3	<3	10			
1,2,4-Trimethylbenzene	μg/l	8	<3	<3	_			
4-Isopropyltoluene	μg/l	10	<3	<3	-			
Naphthalene	μg/l	68	<2	<2	1			
DRO	mg/l	0.351	0.092	< 0.01	0.01			
Mineral Oil	mg/l	< 0.01	< 0.01	< 0.01	0.01			
sVOC	μg/l	ND	ND	ND	_			
Pesticides	μg/l	ND	ND	ND				

The results confirm the presence of an aged leachate in the waste mass. The leachate levels range from 91.68 - 92.95mOD and indicate variable levels associated with localized perching within the waste body. The water levels in the wells immediately outside the landfill ranges from 89.24 - 94.86mOD.

The variations in level again probably relate to local variations in the natural subsoil permeability. The variations in level between the leachate and the surrounding wells do not indicate a direct hydraulic connection between the leachate and the groundwater in the gravels. However, the levels of manganese, chloride and ammonia detected in the wells outside the landfill footprint do indicate that leachate has migrated, albeit at dilute concentrations, into this gravel zone.

#### 3.4 Surface Water Monitoring

A surface water drain flows from the site to the east for 150m and then turns south towards and passes beneath the access road towards a housing development. Where the drain reaches this development it is piped through Tipperary Town and eventually discharges to the River Ara. A surface water sample was collected from the drain approximately 50m downstream of the waste body during the Exploratory Works, but the results were not available for inclusion in the Exploratory Phase Report.

#### 3.4.1 Sampling Methodology

The sampling was carried out by full submergence of the laboratory supplied sample containers into the surface water body where possible. During submergence every effort was made to keep the container steady so as to prevent sediment disturbance. Field measurements of temperature, pH, electrical conductivity and dissolved oxygen were recorded.

The samples were stored to color boxes to maintain sample temperature at approximately 4°C. All the samples were submitted to Jones Environmental Forensics in the UK within 24 hours of sampling. The COC is included in Appendix 3.

#### 3.4.2 Laboratory Analysis

The samples were analysed for a range of organic and inorganic parameters that included indicators of general water quality and leachate contamination. The laboratory methodologies were all ISO/CEN approved or equivalent and, with the exception of ammonia, the method detection limits were all below the relevant guidance limit.

#### 3.4.3 Laboratory Results

The laboratory test report is contained in Appendix 3 and the results are summarised in Table 3.10. The table includes for comparative purposes Environmental Quality Standards (EQS) published by the EPA. The EQS limits are proposed water quality standards and are derived from the EU Directive on Drinking Water Quality 80/778/EEC and the Directive on the Protection of Groundwater against pollution caused by certain dangerous substances 80/66/EEC.

Table 3.10 Surface Water Results

Sample I.D.	Units	SW-1	EQS
рН	pH Units	7.290	4.5-9
Electrical Conductivity	uS/cm	707.000	-
Arsenic	mg/l	0.004	0.025
Boron	mg/l	0.034	-
Cadmium	mg/l	< 0.005	0.0015
Copper	mg/l	< 0.007	0.03
Lead	mg/l	< 0.005	0.0072
Manganese	mg/l	< 0.002	-
Magnesium	mg/l	9.420	-
Mercury	mg/l	< 0.001	0.00007
Nickel	mg/l	< 0.002	0.02
Iron	mg/l	0.182	1
Total Cyanide	mg/l	<0.04.	0.01
Total Chromium	mg/l	<0.0015	0.0047
Zinc	mg/l 👌	N. 20.003	0.1
Sulphate	mg/l	26.99	200
Chloride	mgylecult	28.16	250
Calcium	mg/l	103.40	-
Fluoride	the simg/l	< 0.3	5
Phosphorus	🔊 mg/l	0.07	-
Total Oxidised Nitrogen	mg/l	0.50	No Abnormal change
Total Suspended Solid	mg/l	6.00	-
Total Alkalinity as CaCO3	mg/l	240.00	-
BOD	mg/l	3.00	5
COD	mg/l	23.00	-
Potassium	mg/l	5.44	-
Sodium	mg/l	17.54	-
Ammonia*	mg/l	1.32	0.02
РАН	mg/l	ND	-
VOC	mg/l	ND	-
sVOC	mg/l	ND	-
Pesticides	mg/l	ND	-
DRO	mg/l	< 0.01	-
Mineral Oil	mg/l	< 0.01	-

ND - denotes not detected

With the exception of ammonia (1.7mg/l), all of the parameters were less than the relevant EQS and there is no evidence of leachate contamination. It should be noted however that the sampling took place after a period of very wet weather, when dilution was significant.

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#### 3.5 Landfill Gas Monitoring

Landfill gas monitoring was conducted in all eight monitoring wells (MW-1 to MW-8) on the 23<sup>rd</sup> November, 2<sup>nd</sup> December and on the 10<sup>th</sup> December 2009. The monitoring included the measurement of methane, carbon dioxide, oxygen and atmospheric pressure using a Gas Data LSMx gas analyser. The meter was calibrated before use. The detection limit is 0.1% for methane, carbon dioxide and oxygen.

The results are presented in Table 3.11. The table includes guideline limits taken from the Department of the Environment (DOE) publication on the 'Protection of New Buildings and Occupants from Landfill Gas' (1994).

#### 3.5.1 Wells Inside the Waste Body

MW-1, MW-2 and MW-3 are all located within the waste body. Carbon dioxide and methane were detected in all three wells ranging from 31.5% to 56% for methane, and 1.5% to 16% for carbon dioxide. Oxygen levels ranged from 0.8% to 1.4% for oxygen.

The guidelines stipulate that, where carbon dioxide or methane are present in a landfill at 0.5% v/v and 1% v/v respectively, then housing should not be erected within 50 m of the landfill and private gardens should not be allowed within 10 m. There is one building on site, which was previously used of materials including pipes. This building is no longer in use, has been sealed and will shortly be demolished. There is a halting site located approximately 150m to the south of the site.



#### 3.5.2 Wells Outside the Waste Body

MW-4, MW-5, MW-6, MW-7 and MW-8 are all located outside the waste body. Methane was only detected in one of the wells, MW-8 along the eastern site boundary where the concentrations ranged from 0.8% to 1.3%. Carbon dioxide was detected in all of the wells, with the concentrations ranging from 0% to 5%. The oxygen levels ranged from 2.9% to 22.6%, with the lowest level detected in MW-8.

Monitoning Woll		Methane		C	Carbon Dioxid	le		Oxygen		Bar	ometric Pres	sure
Monitoring weil	23/11/2009	02/12/2009	08/12/2009	23/11/2009	02/12/2009	08/12/2009	23/11/2009	02/12/2009	08/12/2009	23/11/2009	02/12/2009	08/12/2009
MW-1	31.5	53	52	12	15	16	1.4	1.1	1	1002	1001	1002
MW-2	55	55	56	3.6	3.9	4.1	1.3	1.1	1.1	1002	1001	1002
MW-3	35	37.5	38	1.5	3.6	3.7	ther ¥.1	0.8	0.9	1002	1001	1002
MW-4	0	0	0	1.9	2.1	2.519. an	22.3	19.9	18.4	1002	1001	1002
MW-5	0	0	0	1.6	0.9	purpose di	18.1	21.8	21.6	1002	1001	1002
MW-6	0	0	0	1.8	4 ection	<sup>rei 3.6</sup>	21	20.1	20.1	1002	1001	1002
MW-7	0	0	0	0	FORMER	1	22.6	3.7	19.1	1002	1001	1002
MW-8	0.8	1.1	1.3	5	at of 4.5	4.6	2.9	3.6	3.6	1002	1001	1002
DOE Limit (%)		0.5%		Ce	1%			-			-	

 Table 3.11
 Landfill Gas Monitoring Data: November and December 2009

DOE limit not established

#### **3.6 Waste Characterisation**

Waste Characterisation had been undertaken during the Exploratory Phase Site Investigations which were detailed in a separate Exploratory Phase Site Investigation report. As part of the assessment two waste samples were selected to be analysed but results had not been received in time for the completion of that report. The results are incorporated below to complete the characerisation process. The samples were analysed for the parameters set out in the EU Council Decision establishing criteria and procedures for the acceptance of waste at landfills pursuant to Article 16 of and Annex II to Directive 1999/31/EC (Council Decision).

The Council Decision sets threshold limits for a range of inorganic and organic parameters, which define whether a waste is suitable for disposal to an inert, non-hazardous or hazardous waste landfill. Based on field observations it was considered the parameters specified in the Council Decision were appropriate for assessment purposes. However, depending on the test results, additional analyses may be carried out.

The solid samples were tested for Total Organic Carbon (TOC), BTEX (benzene, toluene, ethylbenzene and xylene) Polychlorinated biphenyls (PCB), Mineral Oil and Polycyclic Aromatic Hydrocarbons (PAH). Leachate generated from the waste samples were tested for metals (arsenic, barum, cadmium, chromium, copper, mercury, molybdenum, nickel, lead, antimony, selenium and zinc), chloride, fluoride, soluble sulphate, phenols, dissolved organic carbon (DOC) and total dissolved solids (TDS). The laboratory methodologies were all ISO approved or equivalent and the method detection limits (MDL) were all below the relevant guidance limit.

The full laboratory test report is in Appendix 3 and the results are summarised in Table 3.1. Included in the Fable are the WAC for Inert and Non-Hazardous Waste.

Parameter	Unit	TP-4	TP-15	Inert Landfill	Non- Hazardous Landfill
Antimony	mg/kg	0.03	0.24	0.06	0.7
Arsenic	mg/kg	0.08	0.115	0.5	2
Cadmium	mg/kg	< 0.01	<0.01	0.04	1
Copper	mg/kg	<0.12	<0.12	2	50
Chromium	mg/kg	< 0.02	< 0.02	0.5	10
Lead	mg/kg	<0.1	<0.1	0.5	10
Nickel	mg/kg	< 0.06	0.05	0.4	10
Molybdenum	mg/kg	0.13	0.95	0.5	10
Selenium	mg/kg	< 0.03	< 0.03	0.1	0.5
Zinc	mg/kg	0.05	0.09	4	50
Mercury	mg/kg	< 0.001	< 0.001	0.01	0.2
Barium	mg/kg	0.48	1.92	20	100
Chloride	mg/kg	70	1847	800	15,000
Fluoride	mg/kg	<1	<1	10	150
Sulphate*	mg/kg	503	934	1000*	20,000
Dissolved Organic Carbon	mg/kg	120	190	500	800
Total Dissolved Solids	mg/kg	2,340	5,860	4,000	60,000
Phenols	mg/kg	<1	s¢₽.	1	NE
Total Organic Carbon	mg/kg	2,100	\$2,300	30,000**	NE
Benzene	mg/kg	<0.002	< 0.002	6	NE
Toluene	mg/kg	<0.002	< 0.002	6	NE
Ethylbenzene	mg/kg	< <u>€</u> €0002	< 0.002	6	NE
Total Xylene	mg/kg	. <b>√</b> <0.006	< 0.006	6	NE
PCB Total of 7	mg/kg/1	<0.035	< 0.035	1	NE
Naphthalene	mg/kg	< 0.03	< 0.03	NE	NE
Acenaphthylene	mg/kg	< 0.02	0.09	NE	NE
Acenaphthene	mg/kg	< 0.02	0.06	NE	NE
Fluorene	mg/kg	< 0.02	0.09	NE	NE
Phenanthrene C <sup>or</sup>	mg/kg	0.29	0.76	NE	NE
Anthracene	mg/kg	0.10	0.31	NE	NE
Fluoranthene	mg/kg	0.59	1.96	NE	NE
Pyrene	mg/kg	0.49	1.60	NE	NE
Benzo(a)anthracene	mg/kg	0.40	1.15	NE	NE
Chrysene	mg/kg	0.40	1.18	NE	NE
Benzo(b)+Benzo(k)fluoranthen e	mg/kg	1.09	2.21	NE	NE
Benzo(a)pyrene	mg/kg	0.58	1.57	NE	NE
Indeno(123cd)pyrene	mg/kg	0.46	1.06	NE	NE
Dibenzo(ah)anthracene	mg/kg	0.40	0.61	NE	NE
Benzo(ghi)perylene	mg/kg	0.55	1.09	NE	NE
Coronene	mg/kg	0.19	0.3	NE	NE
Total 17 PAH's	mg/kg	5.54	14	NE	NE
Mineral Oil	mg/kg	<30	<30	500	NE

#### **Table 3.12 Waste Characterisation**

#### **NE - Not Established**

\* - sulphate level exceeding inert waste limit may be considered as complying if the TDS value does not exceed 6,000mg/kg at L/S = 10l/kg.

\*\*-a higher limit may be accepted provided the DOC values of 500mg/kg is achieved

The majority of the parameters were below the Inert Limits; however the levels of antimony, molybdenum, chloride and TDS exceeded the WAC Inert Limit in the samples from TP-15. No significant zones of potentially contaminated or hazardous waste were identified during the site investigations. Based on the site observations during the investigation and the sampling results, the waste is considered to be Non-Hazardous and typical of that found in small scale municipal landfills.

#### 3.7 Re-evaluation of the Geophysical Site Investigation

The Agency review of the Geophysical Site Investigation noted "that a reclassification of the aquifer vulnerability was being considered due to the results of the geophysical survey. Caution should be observed when reclassifying bedrock on resistivity surveys only – it is best practice to also use seismic geophysics for that interpretation"

OCM requested APEX Geoservices, who conducted the survey, to review the findings in response of the EPA's comments. APEX comments are as follows:

The Tier 1 Desk Study Assessment indicated the presence of possible high vulnerability and karstic bedrock aquifer beneath the site. The expected resistivity response on R5 & R6 (resistivity lines run outside the landfill footprint) for high vulnerability and Waulsortian karst linestone would be <u>high resistivity (1000-5000 ohm-m)</u> at relatively shallow depth (5,50m). This was not the case and <u>low resistivity</u> (50-250 ohm-m) was observed to c 22m bgl on both profiles. This resistivity range is typical of South Midlands gravelly clays with occasional gravel pockets.

At c. 22m, there was a distinct transition on both R5 and R6 to slightly higher resistivity material (250-400 ohm-m), which was interpreted as shaley or argillaceous bedrock (probable Athassel dark shaley cherty limestone) rather than the expected cleaner Knockordan or Waulsortian-equivalent limestone. Such variations from the published geological maps would not be unusual in drift covered areas.

Given the strongly contrasting resistivity readings (250 - 400 ohm-m found versus 1000 - 5000 ohm-m typical of the Rkd Aquifer mapped for the area beneath the site) OCM concurs with the Apex Geoservices comments

A deep borehole (MW-7) was installed away from the landfill footprint to try to establish the nature of the bedrock. Dark grey shaley limestone gravels were encountered from approximately 15.5 -20m when the drilling terminated due to the nature of the gravels.

O' Callaghan Moran & Associates

While top of bedrock was not confirmed, the nature of the gravels and difficulty in advancing the drill stem in this hole suggests that the top of bedrock was close. This is supported by the depth to bedrock estimated in the geophysical survey. OCM considers that, based on the nature of the gravels at depth (shaley limestone), and the geophysical interpretation provided by Apex, that the bedrock beneath the site is a shaley limestone and not the Waulsortian karst aquifer.

#### 3.8 Assessment of Surface and Groundwater Pathways for Leachate Migration

#### 3.8.1 Surface Water

Leachate levels recorded in the waste (MW-1 – MW-3) range from 91.68 - 92.95mOD. These levels are all typically more than 0.5m above the natural ground level surrounding the landfill and indicate that the leachate can potentially migrate through the landfill into the Marsh and ultimately into the drain to the east. It is possible therefore that leachate reaches the drain through the ponded marsh area seeping out along the base of the waste. However, based on the existing data the level of impact at such times is very low due to dilution by run-off from the surrounding lands.

No leachate break outs were observed around the margins of the landfill and the marsh area did not look to be significantly impacted as a result of leachate discharge from the waste. Long term monitoring of the marsh and surface water drain is required to determine if surface water is impacted by leachate discharges from the site, particularly during low flow conditions. Monitoring of water quality in the marsh area may also be required though it is likely that the water in this area will be stagnant with low oxygen levels given its topographically low setting.

A groundwater flow direction has been compiled for the site (Figure 3.2). The groundwater table appears to reflect the topography of the natural ground with a low point in the vicinity of MW-8 to the east of the landfill and flow toward this area from all other areas.

The log for MW-4 located 10m from the surface water drain to the east of the landfill well indicates that the subsoil here comprise 7-8m of clay between the gravel/clay zone at depth and the base of the drain where the drain flows from north to south to the east of the landfill. There does not appear to be any direct connection therefore between this clayey gravel zone at depth and the drain. It appears that the marsh area is the outlet for groundwater /leachate within the catchment and that as the marsh area fills up it over flows to the surface water drain. Leachate migration from the waste is occurring from the waste body toward the natural low point in the marsh along the interface between the base of the waste and compacted natural ground where the landfill merges with the marsh area.

It is unclear if the upper gravelly clay zone is just a lense of gravel in the local area but it has been intersected in all the shallow wells. Groundwater level data indicates that currently leachate migration is not occurring through this zone down hydraulic gradient to the east and south toward the River Ara

The presence of a hard dry 1.75m thick clay layer beneath the uppermost gravel zone would most likely prevent leachate migration to the bedrock aquifer. As the available site investigation data indicates that the aquifer is likely to be Ll, the risk posed is less significant than if the aquifer were a regionally important karst bedrock aquifer.

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The public supply well (Tipperary CoOp) is located 1.5 km to the south of the site are uncontaminated, which indicates that leachate contamination in the gravels is not impacting on water quality in the bedrock aquifer.

Comparing the concentrations of (ammonia, chloride and manganese) detected in the groundwater wells with the concentrations in the leachate in the waste indicates that substantial dilution and attenuation is occurring within 5-10m of the landfill mass. It is considered likely there that further dilution and possible natural attenuation is occurring in the marsh area prior to discharge out of the catchment in the drain.

#### 3.9 Assessment of Landfill Gas Pathway

The monitoring in the waste body (MW-1, 2 and 3) indicates that methane and carbon dioxide are still being generated at significant levels. The monitoring in the perimeter wells identified carbon dioxide levels ranging from 2.5 to 4.6%, however methane was only detected at one location, MW-8 located 10m to the east of the of the waste.

While the levels of  $CO_2$  are above the DOE limits they are not very elevated and could at a distance from the site (MW-4) in part be naturally occurring. Methane is most likely associated with the presence of landfill gas emanating from the waste body. Because of the presence of the marsh and ponding water it was not possible to install monitoring wells to the north of the landfill. It is possible that some landfill gas migration may also be occurring into the marsh area. However the marsh land probably acts as a buffer between the landfill and the reclaimed lands to the north of the site, allowing ventilation to atmosphere or dissolution into the ponded water during the winter months.

The on-site building is no longer used and it is planned to demolish the structure in the near future. The risk to on-site users has therefore been eliminated.

The lands north and northwest, approximately 200 - 300m away, have been reclaimed with construction demolition waste as part of the planned future development for residential and commercial purposes. Currently there is a halting site located 150m to the south of the site. There are no other residential dwellings within 250m of the site. It is also possible that in the future the lands surrounding the site could be developed for residential and/or commercial purposes. Long term monitoring for landfill gas will be required around the landfill to ensure that any risk posed to areas proposed for development in the future can be mitigated in advance.

# 4. REVISION OF TIER 2 RISK ASSESSMENT & CONCEPTUAL SITE MODEL

#### 4.1 Revised Conceptual Site Model

A revised conceptual Site Model is presented on Figure 4.1 below. This model illustrates the presence of low –to moderate permeability boulder clay and gravel which are in turn underlain by layer of low permeability hard clays. Beneath the clay layer is a layer of gravels which in turn appears to overly the shaley limestone Ll aquifer beneath the site. The landfill appears to be located at the low point in a local catchment where both groundwater and surfacewater discharge via a Marsh to the drain to the east of the site.

The leachate, while present in the shallow ground water is likely to be contained above the deeper clay layer. Direct discharge to drain via the groundwater pathway is not considered to occur. This appears to occur via the Marsh which in turn discharges slowly to the drain as water levels rise in the winteroperiod. The marsh conditions more than likely result in some natural attenuation of the leachate prior to discharge of water from the marsh to the stream.





#### 4.2 Revised Risk Assessment

OCM has modified the original risk assessment on the basis of the findings of the Exploratory Phase and the changes are highlighted in blue.

Table 6
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1 401			
Ref	Source	Score	Rational
1a	Leachate	7	<5 hectares
			<ul> <li>Waste likely to be both municipal &amp; industrial</li> </ul>
1b	Gas	7	<5 hectares
			• Highest rating given as proportion of municipal:
			industrial waste is not known.

Table 7

Ref	Pathways	Score	Rational
2a	Groundwater	2	• GSI data states that the site is rated as having high
	vulnerability		vulnerability.
2b	Groundwater flow	1	• Bedrock was originally considered to be karst. The
	regime		drilling data and geophysical survey indicates that
			bedrock is soot karst and is likely to be Shaley
			Limestone i.e Ll Aquifer. Score reduces from 5 to
			1. 5 X X X
2c	Surface water drainage	2	• <b>Land</b> fill is reportedly connected to town surface
			water drainage system
2d	Landfill gas lateral	3	Residences not currently within 250m of site, but
	migration	THE	will be within 5 years
		to Ar	<ul> <li>Karst bedrock</li> </ul>
2e	Landfill gas vertical	. 80 <sup>000</sup>	• Building on site not occupied and will be removed
	migration	Sent	risk score reduces from $5-0$
	CÔ	<u>r</u>	

#### Table 8

Ref	Receptors	Score	Rational
3a	Human presence (leachate)	2	<ul> <li>Currently no houses within 250m, there will be within 5 years</li> <li>Note: All houses will be served by public water</li> </ul>
3b	Protected areas	1	<ul> <li>No protected areas within 1 km of site</li> <li>The marsh area has been considered as an undesignated GWDTE, precautionary approach.</li> <li>No consultation with the NPWS has taken place.</li> </ul>
3c	Aquifer category	5	<ul> <li>Locally Important L1 aquifer underlies the site, score reduces from 5 to 3</li> </ul>
3d	Public water supply	3	<ul> <li>Public water supply is greater than 1km away (Cordangan)</li> <li>Karst bedrock – but different geological formation</li> <li>Precautionary approach assumed</li> </ul>
3e	Surface water bodies	3	<ul> <li>Surface water drain within 50m of site boundary</li> </ul>
3f	Human presence (gas)	0	<ul> <li>Building on site unoccupied and to be removed score reduces from 5 to 0</li> </ul>

The risk assessment revised after the Detailed Site Investigations indicates that the site remains High risk for leachate because the pathway from the marsh to the drain appears viable. Further monitoring is required to establish if impacts on surface water quality is occurring during low flow conditions.

Landfill Gas Risk remains Moderate because of detection of carbon dioxide in the perimeter wells and methane in one well. The risk posed to receptors cannot be eliminated and needs to be assessed as part of a longer term monitoring programme.

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Note: The table below represents the Tier 1 Risk rating for this site. SPR1 to 9 represent the leachate risk scores. SPR10 & 11 represent Landfill Gas Risk. The migration pathways are colour coded as follows:

 Groundwater &
 Groundwater &

Surface Water	Groundwater only	Surface water only	Lateral & Vertical	
				N 10
Ca	lculator	SPR Values	Maximum Score	Normalised Score
SPR1	1a x (2a + 2b + 2c) x 3e	105	300	35.00%
SPR2	1a x (2a + 2b + 2c) x 3b	0	300	0.00%
SPR3	1a x (2a + 2b) x 3a	42	240	17.50%
SPR4	1a x (2a + 2b) x 3b	0	1 <sup>5<sup>e.</sup> 240</sup>	0.00%
SPR5	1a x (2a + 2b) x 3c	63	other 400	15.75%
SPR6	1a x (2a + 2b) x 3d	0 0113	560 start	0.00%
SPR7	1a x (2a + 2b) x 3e	63 0 <sup>56°</sup> d <sup>10</sup>	240	26.25%
SPR8	1a x 2c x 3e	42 purpequite	60	70.00%
SPR9	1a x 2c x 3b	0 citomert	60	0.00%
SPR10	1b x 2d x 3f	2,159,110 m	150	14.00%
SPR11	1b x 2e x 3f	This linkage is not present du	e to no receptor above the source	
Overal	l Risk Score	\$105		70.00%
		n <sup>sent</sup>		A
		C O*		

Risk Classification	Range of Risk Scores
Highest Risk (Class A)	Greater than 70 for any individual SPR linkage
Moderate Risk (Class B)	40-70 for any individual SPR linkage
Lowest Risk (Class C)	Less than 40 for any individual SPR linkage

<b>KISK Classification</b>
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### 5. CONCLUSIONS AND RECOMMENDATIONS

#### 5.1 Conclusions

The Detailed Site investigation indicates that the site currently a Class A High Risk Site based on the risk posed to the surface water system.

However, impact on the surface water quality in the drain is currently low with only ammonia exceeding EQS limits. This may in part be due to natural attenuation within the marsh and very high rainfall levels which are attenuating the impact on the surface water system.

It is possible that some leachate migration is occurring from the site toward the marsh wetland area and into the drain, which ultimately discharges to the River Ara several kilometres downstream of the site.

Based on the groundwater flow direction data the groundwater in the catchment is moving to a low point in the vicinity of the marsh. It is likely that the marsh is therefore the local discharge for groundwater.

Significant dilution of leachate appears to be occurring between the landfill and the groundwater based observation of the substantial reduction in Manganese, Chloride and Ammonia levels seen between the leachate in the waste and the external monitoring wells located within 5-10m of the landfill.

Water quality in the public groundwater abstraction well located 1.4km down hydraulic gradient of the site is of good quality which supports the assumption that the clay layer beneath the gravels and above the bedrock inhibits the vertical migration of groundwater.

Some remedial measures may be required to minimize the risk posed by leachate and landfill gas to off-site receptors but further monitoring of landfill gas, surface water quality and groundwater quality over a longer time period will be required to establish the extent of remediation required.

#### 5.2 Recommendations

#### 5.2.1 Leachate Risk

OCM recommend that the more recently deposited waste material deposited at the landfill be capped with low permeability soils/subsoils to minimise the infiltration of rainfall to the waste.

#### 5.2.1.1 Surface Water

OCM recommend that monitoring of the surface water in the marsh area and in the drain be undertaken during lower flow conditions to establish if leachate is migrating into the marsh area and/or into the stream. Monitoring should be undertaken for an initial period of 6 months from April to September on a bimonthly basis (once every two months).

#### 5.2.1.2 Groundwater

Following capping of the site OCM recommend that monitoring of the groundwater be undertaken to establish the effectiveness of the capping programme in reducing the generation of leachate beneath the site. Monitoring should be undertaken at least bi-annually for this purpose.

# 5.2.2 Landfill Gas

OCM recommend that when the water levels receed that landfill gas monitoring wells be installed to the north, north east and northwest of the marsh area.

OCM recommend that landfill gas monitoring be undertaken particularly in the southern section of the site (MW5, 6 and 7) at monthly intervals to assess the risk of off-site migration toward the halting site, the residential area within 250m of the site.

OCM recommend that all wells be monitored at least annually to assess landfill gas levels. In the event that development occurs within 250m of the site boundary the more frequent monitoring would be required.

If monitoring shows that landfill gas levels increase as a result of capping OCM recommend that mitigation measures be developed to reduce the landfill gas levels. Such measures may include the installation of additional landfill gas wells within the site to ventilate the site.



December 2009 (BS/SM)


Client: South Tipperary Co. Co.

Project: 09-188-01

Borehole Depth: MW-01

SWL (m): 4.34m

Depth (m)	Lithology Description	Lithology	Well Construction Details
-1 0 1 2 3 4 5 6 7 8 9 10 11 12 13 14	FURVASE         Full comprised of black sandy gravelly Clay with flactics, timber, glass and papers.         Water strikes at 1.3m, 6.4m, 10.85m.         Grave compression of the sandy gravelly clay with the sandy gravelly clay with the sandy gravelly clay.         Conserve compression of the sandy gravelly clay.         Provide compression of the sandy gravelly clay.         Provide compression of the sandy gravelly clay.		Slotted 50mm HDPE Pipe Bentonite Plug Bentonite Plug Gas Cap Bentonite Plug Filter Pack Blain 50mm HDPE Pipe Steel Headworks
Drilli	ing Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 16/11/2009	She	et: 1 of 1



Project: 09-188-01

Borehole Depth: MW-02

Client: South Tipperary Co. Co. SWL (m): 5.8m

Depth (m)	Lithology Description	Lithology	Well Construction Details
-1- 0- 1- 2- 3- 4- 5- 6- 7- 8-	Ground Surface Fill/Waste Fill comprised of black sandy gravelly Clay with plactics, timber, glass and papers. Water strikes at 5.85m and 8.25m. Water strikes at 5.85m and 8.25m. Water strikes at 5.85m and 8.25m.		PE Pipe Gas Cap Gas Ca
9- 10- 11- 12-	Clay		Slotted 50mm HD
	Very stiff brown sandy gravelly Clay.		
Drilli	ng Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 17/11/2009	She	et: 1 of 1



Project: 09-188-01

Borehole Depth: MW-03

Client: South Tipperary Co. Co. SWL (m): 6.91m

Depth (m)	Lithology Description	Lithology	Well Construction Details
-1- 0- 1- 2- 3- 4- 5- 6- 7- 8- 9- 10- 11- 11-	FII/Wase         Fill comprised of black sandy gravelly Clay with plactics, timber, glass and papers.         Water strike at 5.85m.         Compression of the strike at 5.85m.     <		Slotted 50mm HDPE Pipe Gas Cap Bentonite Plug Steel Headworks Carvel Filter Pack Steel Headworks
Drilli	ng Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 17/11/2009	She	et: 1 of 1



Project: 09-188-01

Borehole Depth: MW-04

Client: South Tipperary Co. Co. SWL (m): 0.2m

Depth (m)	Lithology Description	Lithology	Well Construction Details
	Crewel Clay Very soft brown Clay. Clay Very stiff brown sandy gravelly Clay with occassional boulders. Gravels are subangular dark grey imestone. Conserver of the subangular dark grey limestone. Conserver of the subangular dark grey limestone. Groundwater strike at 8.8m.		Slotted 50mm HDPE Pipe Slotted 50mm HDPE Pipe Steel Headworks Cavel Filter Pack Bentonite Plug Steel Headworks Steel Headworks
-			
Drilli	ng Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 18/11/2009	She	et: 1 of 1



Client: South Tipperary Co. Co.

Project: 09-188-01

Borehole Depth: MW-05

SWL (m): 1.85m

Depth (m)	Lithology Description	Lithology	Well Construction Details
-1- $0 1 2 3 4 5 6 7 8 9 10 10-$	<section-header></section-header>		Slotted 50mm HDPE Pipe Bentonite Plug Plain 50mm HDPE Pipe Steel Headworks Steel Headworks
Drilli	ng Contractor:	Hole	e Size:
Drill Method: Air Rotary Drill Date: 18/11/2009			logist: B. Sexton et: 1 of 1



Project: 09-188-01

Borehole Depth: MW-06

Client: South Tipperary Co. Co. SWL (m): 0.7m

Depth (m)	Lithology Description	Lithology	Well Construction Details
	Clay Very soft brown Clay. Clay Very stiff brown sandy gravelly Clay with occassional boulders. Gravels are subangular dark grey limestone.		Slotted 50mm HDPE Pipe Bentonite Plug Bentonite Plug Plain 50mm HDPE Pipe Gravel Filter Pack Steel Headworks
11-	Groundwater strike at 9.5m		
Drill	ing Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geol	logist: B. Sexton
Drill	Date: 19/11/2009	Shee	et: 1 of 1



# Borehole I.D. MW-07 Revised

Project: 09-188-01

Client: South Tipperary Co. Co.

Borehole Depth: MW-07 Revised

SWL (m): 6.35m

			-
Depth (m)	Lithology Description	Well Construction Details	
-1-			
-	Ground Surface		Cap
-	<i>Clay</i> Clay with minor amounts of concrete.		adwo
- - 1- - -	<i>Clay</i> Brown very stiff sandy gravelly Clay. Gravels are subangular dark grey limestone.		Pluite Plui
2-	1005 COLUMN AND OL		Ben Ben Som HD
3-	ction per rectu		
- - - 4-	Sand Brown medium sand.		
5-	Gravelly Clay Brown very stiff sandy gravely Clay. Gravels are subangular dark grey limestone.		
6			vel Fiiter F
7-			Gray Gray
- - 8- -	<b>Gravel</b> Clay rich dark grey limestone Gravels. Gravels are subangular to subrounded.		Slotted
- - 9 -	Groundwater strike at 8.85m.		
- 10-			
Drilli	ng Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 19/11/2009	She	et: 1 of 2



# Borehole I.D. MW-07 Revised

Project: 09-188-01

Borehole Depth: MW-07 Revised

Client: South Tipperary Co. Co. SWL (m): 6.35m

Depth (m)	Lithology Description	Lithology	Well Construction Details
	Sand and Gravel Dark grey limestone sand and gravel. Gravelly Clay Brown very stiff sandy gravelly Clay, Gravels are subangular dark grey limestone. For stiff		vith Cement Grout
	Gravel Clay rich drak grey limestone Gravels. Gravels are subangular to subrounded.		ole Backfilled
- - 17- -	Groundwater strike at 15.75m.		Broeh
- - 19- -			
20-			
21-			
Drilli	ing Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geo	logist: B. Sexton
Drill	Date: 19/11/2009	She	et: 2 of 2



Project: 09-188-01

Borehole Depth: MW-08

Client: South Tipperary Co. Co. SWL (m): 3.94m

Depth (m)	Lithology Description	Lithology	Well Construction Details
-1 - 0 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	Peaty Clay. Peaty Clay. Clay We stiff brown sandy gravelly Clay. Gravels are subangular dark grey limestone. Occasional limestone boulders. Conserver of the provide of th		Sotted 50mm HDPE Pipe Bentonite Plug Finite Strike 8.95m Mater Strike 8.95m Cavel Filter Pack Branch Plain 50mm HDPE Pipe Steel Headworks
11	Groundwater strike at 8.95m.		
Drilli	ing Contractor:	Hole	e Size:
Drill	Method: Air Rotary	Geol	logist: B. Sexton
Drill	Date: 20/11/2009	Shee	et: 1 of 1



December 2009 (BS/SM)



## STANDARD OPERATING PROCEDURE

### SURFACE WATER SAMPLING

The primary objective of surface water sampling is to evaluate the chemical quality of a water body. The purpose of this procedure is to ensure that representative samples of surface water are collected and documented using consistent methods to ensure sample integrity. Surface water grab samples may be collected from rivers, streams, lakes and wetlands. In cases where the depth of the surface water body prevents sampling from the banks of the water body, sampling from, a boat may be required.

#### 1.0 SAMPLING PROCEDURES

#### 1) 1.1 **Equipment** Needed

- 25e5 only any other use Personal protective clothing and equipment as required in the site-specific risk ٠ assessment.
- Decontamination equipment and supplies if known contaminated site.
- Temperature probe EC meter, pH meter, dissolved oxygen meter.
- Appropriate sample containers (some will be pre-preserved), labels and chain of . custody documentation.
- Field logbook.
- Hard plastic cooler with ice pack.

#### 1.2 **Field Parameter Measurement**

Measurements of field parameters of pH, temperature and electrical conductivity are made during sampling. Note visual (colour, turbidity) and odour (e.g hydrocarbon, hydrogen sulphide) characteristics in the field logbook.

## 1.3 Collection of Water Samples

All samples for chemical analysis will be placed in laboratory prepared bottles. The types of sample containers and preservative required for each type of analysis are described in the workplan. If required, preservatives will be placed in the sample containers prior to collecting the samples.

The following procedure will be used -

- 1) Slowly submerge unpreserved one-liter amber glass or plastic-capped bottles completely into the water. Open and fill bottle from below the water surface. If wading is required, approach the sample site from downstream and do not enter the actual sample area. Do not disturb bottom sediments. Open-end of the bottle should be pointed at approximately 90° to the upstream direction, in undisturbed gently flowing water. This procedure will be performed to minimize the effects due to high turbulence and aeration, or if surface scum is prevalent.
- 2) Collect a sufficient volume of water to fill all sample containers.
- 3) For VOC analysis. Pour the samples slowly into the laboratory prepared 40 ml glass vial. Overfill each vial slightly to eliminate air bubbles, a convex meniscus should be present at the top of the vial. Ensure that the fellon liner of the septum cap is facing inward and that no bubbles are entrapped. After capping securely, turn bottle upsidedown, tap it against your other hand, and observe sample water for bubbles. If bubbles are observed, remove the cap, overfill the vial and reseal. Repeat this step for each vial until the samples with no bubbles are obtained.
- 4) Obtain the semi-volatile compound/pesticides/PCBs sample(s) by transferring the water to a laboratory prepared 1000 ml amber glass bottle with Teflon-lined cap. Fill the bottle to the bottom of the neck and follow steps 4, 5 and 6 above.
- 5) Dissolved metals (if necessary) may require filtering the sample water through a .45 micron filter. The water is collected in a 1 litre, unpreserved, plastic or glass bottle with HNO<sub>3</sub> preservative. Filtering must be done within 15 minutes of sample collection.
- 6) Obtain the total metals sample by directly transferring the water into a laboratory prepared 1000 ml plastic or glass bottle with HNO<sub>3</sub> preservative. Ensure the pH of the metals sampled is less than 2 by pouring off an aliquot in a clean jar and testing for pH using litmus paper.
- 7) Collect and prepare Field QA/QC samples in accordance with separate SOP.
- 8) Place a label on the container and enter the following information: -

Client/Site Name Date Collected Time Collected Analysis

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Preservative Sample Identification Number

- 9) Place custody seals on the container caps. As soon as possible, place sample containers in a cooler with ice and maintain at 4°C. Surround the bottles with packaging.
- 10) Record pertinent information in the field logbook and on the Field Data Sheet for Sampling Location. Complete chain-of-custody form, place in cooler and seal and label the cooler.
- 11) Be sure to record all data required on the Field Data Sheet or Sampling Location and appropriate entries into the field logbook.
- 12) Decontaminate all sampling equipment according to procedure.

END.

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	on purpose ined	
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## STANDARD OPERATING PROCEDURE

## GROUNDWATER SAMPLING

The primary objective of groundwater sampling is to establish groundwater quality and evaluate whether the potential contaminant sources at a site have impacted the groundwater in the underlying aquifer. The additional objective is to measure hydraulic gradient, or slope, of the water table to evaluate the direction of groundwater flow.

The purpose of this procedure is to ensure that representative samples of groundwater are collected and documented using consistent methods to ensure sample integrity.

## 1.0

### 1.1

Well Operating and Purging Procedures All groundwater sampling will be conducted after the installed and developed wells have been allowed to equilibrate for at least 2 to 3 days. A Field Data Sheet for Well Sampling will be FOTIN completed for each well.

Groundwater sampling teams will use to following procedure for approaching, opening, purging and sampling all wells, unless directed otherwise by a site specific workplan.

- Prior to placing any equipment into the well, decontaminate the sampling equipment 1) according to standard decontamination protocol.
- 2) Ensure you have a working FID/PID, a well key, and a depth-to-water meter.
- 3) Unlock and open the well cap just enough to insert the probe of the PID/FID. Take and record a reading. A decision to upgrade PPE may be necessary based on the FID/PID readings in the breathing zone.
- 4) Where practical, the surface water column will be visually examined for the presence of hydrocarbons, if present or suspected, the thickness of the hydrocarbon layer will be measured using an oil/water interface probe prior to taking the depth-to-water measurement.
- Insert the water level probe into the well and measure and record the static water level 5) to the nearest 0.01 m with respect to the established survey point on top of the well casing.

- 6) Decontaminate the water level probe with DDI water (Do not rinse with any solvents unless product was encountered).
- 7) Calculate and record the minimum volume of water to be purged according to the following conversion factors: -

1 well volume	=	water column in metres x litres/linear metre
50mm casing	=	2.0 LPM
100mm casing	=	8.1 LPM
150mm casing	=	18.2 LPM
200mm casing	=	32.4 LPM

- 8) Purge the well of at least 3 casing volumes by pumping or bailing with a decontaminated submersible pump or PVC bailer equipped with a bottom filling check valve (if the purge volume is low, generally less than 100 litres, the sampling team might find it more efficient to purge with a bailer than a pump). Use a graduated bucket to track the amount of water removed from the well. Periodically determine the pH, temperature and specific conductance of the purged water. Continue purging until the well has been completely evacuated or until the pHe and specific conductance measurements have stabilised for at least one well will be sampled as soon as practical once they recover sufficiently.
- 9) Dispose of purge water collected in the graduated bucket by pouring onto the ground at a distance of 50 to 60 metres from the vicinity of the well. If the water is known or suspected to be significantly contaminated, it may be necessary to store the purge water in a secure container, such as a drum pending proper disposal.
- 10) Be aware and record any unasual occurrence during purging such as cascading (a shallow water entry zone that trickles into the borehole).

### 1.2 Field Parameter Measurement

Measurements of field parameters of pH, temperature and electrical conductivity are collected and organic vapour screening is conducted while the well is purged. To facilitate the collection of basic field parameters, the field team needs to: -

- Purge three well volumes of water from the well and measure field parameters for each well volume removed.
- Collection of water samples should take place after stabilisation of the following parameters: -
  - Temperature  $^+/- 1^{\circ}C$
  - pH (meter or paper)  $^+/$  0.2 units
  - Specific conductivity <sup>+</sup>/- 5%

- If the aforementioned parameters do not stabilise within three purge volumes, the well will be purged up to a maximum of six borehole volumes unless two consecutive sets of stabilised parameters are obtained.
- Note any observations in the field logbook.

### 1.3 Collection of Water Samples

All samples or chemical analysis will be placed in laboratory prepared bottles. The types of sample containers and preservative required for each type of analysis are described in the workplan. If required, preservatives will be placed in the sample containers prior to collecting the samples.

The following procedure will be used to sample a well: -

- 1) After the well has been purged and allowed to recover, sample the well using a properly decontaminated or dedicated disposable bailer. Gently lower the bailer into the water column. Allow the bailer to sink and fill with a minimum of surface disturbance.
- 2) Slowly raise the bailer out of the well. Do not allow the bailer line to contact the ground, either by coiling it on a clean plastic sheet or by looping it from arm to arm as the line is extracted from the well.
- 3) Samples will be collected for VOCs analysis immediately after purging is complete and before other samples are collected. Four the samples slowly into the laboratory prepared 40 ml glass vial. Overfillerach vial slightly to eliminate air bubbles, a convex meniscus should be present at the top of the vial. Ensure that the Teflon liner of the septum cap is facing inward and that no bubbles are entrapped. After capping securely, turn bottle upside-down, tap it against your other hand, and observe sample water for bubbles. If bubbles are observed, remove the cap, overfill the vial and reseal. Repeat this step for each vial until the samples with no bubbles are obtained.
- 4) Place a label on the container and enter the following information: -

Client/Site Name Date Collected Time Collected Analysis Preservative Sample Identification Number

- 5) Record pertinent information in the field logbook and on the Field Data Sheet for Well Sampling. Complete chain-of-custody form.
- 6) Place custody seals on the container caps. As soon as possible, place sample containers in a cooler with ice packs and maintain at 4°C until extraction. Surround the bottles with appropriate packaging.

- 7) Obtain the semi-volatile compound/pesticides/PCBs sample(s) by transferring the water to a laboratory prepared 1000 ml amber glass bottle with Teflon-lined cap. Fill the bottle to the bottom of the neck and follow steps 4, 5 and 6 above.
- 8) Dissolved metals (if necessary) requires the team to filter the sample water through a .45 micron filter. The water is collected in a 1 litre, unpreserved, plastic or glass bottle with HNO<sub>3</sub> preservative. Filtering must be done within 15 minutes of sample collection.
- 9) Obtain the total metals sample by directly transferring the water from the bailer into a laboratory prepared 1000 ml plastic or glass bottle with HNO<sub>3</sub> preservative.
- 10) Be sure the pH of the metals sampled is less than 2 by pouring off an aliquot in a clean jar and testing for pH using litmus paper. Dispose of this water and rinse the jar.
- 11) Collect and prepare Field QA/QC samples in accordance with separate SOP.
- 12) Be sure to record all data required on the Field Data Sheet or Well Sampling and appropriate entries into the field logbook.
- 13) Secure the well cap and replace the locking cover.
- 14) Decontaminate all sampling equipment according to procedure.
- 15) Decontaminate submersible pumps as follows.

Scrub pump and cord in a table of appropriate detergent and potable water Pump at least 80 litres of soapy water through pump Rinse with potable water Pump at least 80 litres of rinse water through the pump Rinse with DI water before lowering pump into the next well.

END,



## STANDARD OPERATING PROCEDURE

## LANDFILL GAS MONITORING

The primary objective of landfill gas monitoring is to assess if gas generation would be likely to give rise to a risk to human health or to the environment. It also helps determine trends in gas generation and migration and evaluates the effectiveness of any in-situ gas control measures. The purpose of this procedure is to ensure that representative measurements of landfill gas are collected using appropriate safety procedures.

## **1.0 SAMPLING PROCEDURES**

All landfill gas monitoring equipment used will be certified intrinsically safe. All landfill gas monitoring equipment shall be regularly calibrated, and serviced according to the manufacturer's specification.

The following procedure will be used for monitoring of landfill gas levels in all monitoring boreholes, unless directed otherwise.

- 1) On arrival at the site, test the equipment in accordance with manufacturer's recommendations and record the ambient gas concentrations, atmospheric pressure and temperature in a field notebook. This ensures the gas analyser chamber is purged prior to monitoring. Record the wind speed and direction and other weather conditions.
- 2) Unlock the borehole cover. Examine the appearance of the standpipe, cap and gas valve and note any damage or changes since previous recordings. Record any visible (steam), audible or olfactory signs of gas migration. Record the ground conditions (e.g. dry, wet, frozen, compacted, loose etc). If signs of gas migration are noted, measurement of gas concentrations should be made around the standpipe to ensure there are no dangerous accumulations of gas.
- 3) If the standpipe is fitted with a gas valve, switch on the gas analyser and securely connect the gas analyser inlet port to the gas sample valve via the inlet tube. Open the gas valve and switch on the analyser pump. Run the pump for sufficient time to remove a representative sample from the borehole. Turn the pump off.
- 4) Record methane  $(CH_4)$ , carbon dioxide  $(CO_2)$  and oxygen  $(O_2)$  peaks and steady concentrations.
- 5) Record atmospheric pressure (mb) and temperature (°C).

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- 6) When measurements are completed, the gas sample valve must be closed and the analyser disconnected.
- 7) A measurement of the depth to water in the borehole should be undertaken after completion of all gas measurements. Insert the water level probe into the well and measure and record the static water level to the nearest 0.01 m with respect to the established survey point on top of the well casing.
- 8) Be sure to record all data required in the field log book.
- 9) Secure the well cap and replace the locking cover.
- 10) Briefly run the pump on the gas analyser to purge the analyser chamber with ambient air before proceeding to the next monitoring location.

END.



# **APPENDIX 3**

Laboratory Reports and Chain of Custody Documentation

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December 2009 (BS/SM)

# Jones Environmental Laboratory



Unit 3 Deeside Point Zone 3 Deeside Industrial Park Deeside CH5 2UA

### Tel: +44 (0) 1244 833780 Fax: +44 (0) 1244 833781



Five soil and one water samples were received for analysis on 5th November 2009 which was completed on 18th November 2009. Please find attached our Test Report which should be read with notes at the end of the report and should include all sections if reproduced. All interpretations and opinions are outside the scope of any accreditation, and all results relate only to samples supplied. All analysis is carried out on as received samples and reported on a dry weight basis unless stated otherwise. Results are not surrogate corrected.

J W Farrell- Jones CChem FRSC Chartered Chemist

### Jones Environmental Laboratory

Client Name:	O'Callagh	nan Morar	n & Associ	ates			Report :		Solids				
Reference:	09-188-0	1											
Location:	Tipperary	, Town La	ndfill				Solids: V=60g VOC jar, J=250g glass jar, T=plastic tub						
Contact:	Barry Sex	kton											
JE Job No.:	09/3607												
J F Sample No.	1-2	3-4	5-6	7-8	9-10								
Sample ID		три	TP 11	TD 12	TP 15								
Sample ID	16-1	16-4	16-11	11-12	11-15								
Depth	-	-	-	-	-								
COC No / misc											Please se	e attached	notes for all
Containers	JT	JΤ	JΤ	JΤ	JΤ						abbreviations and acronyms		acronyms
Sample Date	03/11/09	03/11/09	03/11/09	03/11/09	03/11/09								
Sample Type	Soil	Soil	Soil	Soil	Soil								
Batch Number	1	1	1	1	1								Method
Date of Receipt	05//11/09	05//11/09	05//11/09	05//11/09	05//11/09						LOD	Units	No.
DRO/EPH (C8-40)	~	397	~	~	2192						<30	mg/ kg	TM5/PM8
Mineral Oil (interpretation &	~	<30	~	~	<30						<30	ma/ ka	TM5/PM8
calculation)		-00									-00	ing/ kg	
Benzene #	~	<2	~	~	<2						<2	µg/ kg	TM2/PM7
Toluene #	~	<2	~	~	<2						<2	µg/ kg	TM2/PM7
Ethyl benzene #	~	<2	~	~	<2						<2	µg/ kg	TM2/PM7
m/p-Xylene #	~	<4	~	~	<4						<4	µg/ kg	TM2/PM7
o-Xylene #	~	<2	~	~	<2			24			<2	µg/ kg	TM2/PM7
Total BTEX #	~	<12	~	~	<12			150			<12	µg/ kg	TM2/PM7
MTBE #	~	<4	~	~	<4			ther			<4	µg/ кд	TM2/PM7
PCB 28#	~	<5	~	~	<5		119. 2019				<5	ua/ka	TM17/PM8
PCB 52#	~	<5	~	~	<5	, er	ortor				<5 <5	ua/ ka	TM17/PM8
PCB 101#	~	<5	~	~	<5	1205	rea				<5	µg/ kg	TM17/PM8
PCB 118#	~	<5	~	~	<5	n Puredu					<5	µg/ kg	TM17/PM8
PCB 138#	~	<5	~	~	فن 5>	other					<5	µg/ kg	TM17/PM8
PCB 153#	~	<5	~	~	. ASP AL	or					<5	µg/ kg	TM17/PM8
PCB 180#	~	<5	~	~	or 59.81						<5	µg/ kg	TM17/PM8
Total 7 PCBs#	~	<35	~	~	COR35						<35	µg/ kg	TM17/PM8
T		0.1		- nto	40.0						-0.0	0/	TN004
100 #	~	2.1	~	CORSE	12.3						<0.2	70	T WIUZ T
% Dry Matter	~	79.9	~	~	64.3						<0.1	%	PM4
	sample	ID	Depth				E	PH/DRO Int	erpretations				
Interpretation	3-4	TP-4	-				Nat	urally Occurr	ing Compour	ds			
Interpretation	9-10	TP-15	-			1	Nat	urally Occurr	ing Compour	ds			

#### Jones Environmental Laboratory Client Name: O'Callaghan Moran & Associates

Client Name:	O'Callaghan Moran & As
Reference:	09-188-01
Location:	Tipperary Town Landfill
Contact:	Barry Sexton
JE Job No.:	09/3607

Report :

Solids: V=60g VOC jar, J=250g glass jar, T=plastic tub

Solids

				-								
J E Sample No.	1-2	3-4	5-6	7-8	9-10							
Sample ID	TP-1	TP-4	TP-11	TP-12	TP-15							
Depth	-	-	-	-	-							
COC No / misc										Please se	e attached r	notes for all
Containers	JΤ	JT	JT	JΤ	JT					abbrevi	ations and a	acronyms
Sample Date	03/11/09	03/11/09	03/11/09	03/11/09	03/11/09							
Sample Type	Soil	Soil	Soil	Soil	Soil							
Batch Number	1	1	1	1	1					LOD	Units	Method
Date of Receipt	05//11/09	05//11/09	05//11/09	05//11/09	05//11/09				 	_		No.
PAH 6 Total												
Fluoranthene #	~	0.59	~	~	1.96					<0.02	mg/kg	TM4/PM8
Benzo(bk)fluoranthene <sup>#</sup>	~	1.09	~	~	2.21					<0.04	mg/kg	TM4/PM8
Benzo(a)pyrene <sup>#</sup>	~	0.58	~	~	1.57					<0.02	mg/kg	TM4/PM8
Indeno(123cd)pyrene #	~	0.46	~	~	1.06					<0.02	mg/kg	TM4/PM8
Benzo(ghi)perylene <sup>#</sup>	~	0.55	~	~	1.09					<0.04	mg/kg	TM4/PM8
PAH 6 Total	~	3.27	~	~	7.89					<0.07	mg/kg	TM4/PM8
PAH 16												
Naphthalene #	~	<0.03	~	~	<0.03					<0.03	mg/kg	TM4/PM8
Acenaphthylene	~	<0.02	~	~	0.09					<0.02	mg/kg	TM4/PM8
Acenaphthene #	~	<0.02	~	~	0.06					<0.02	mg/kg	TM4/PM8
Fluorene <sup>#</sup>	~	<0.02	~	~	0.09			<u>ر</u> و.		<0.02	mg/kg	TM4/PM8
Phenanthrene <sup>#</sup>	~	0.29	~	~	0.76			of US		<0.02	mg/kg	TM4/PM8
Anthracene #	~	0.10	~	~	0.31			offic		<0.02	mg/kg	TM4/PM8
Fluoranthene #	~	0.59	~	~	1.96		aly and			<0.02	mg/kg	TM4/PM8
Pyrene #	~	0.49	~	~	1.60	e <sup>c</sup>	ortor			<0.03	mg/kg	TM4/PM8
Benz(a)anthracene#	~	0.40	~	~	1.15		eo			<0.02	mg/kg	TM4/PM8
Chrysene #	~	0.40	~	~	1.18	Direch				<0.04	mg/kg	TM4/PM8
Benzo(bk)fluoranthene#	~	1.09	~	~	2.21	offert				<0.04	mg/kg	TM4/PM8
Benzo(a)pyrene <sup>#</sup>	~	0.58	~	~	1.570	OWNE				<0.02	mg/kg	TM4/PM8
Indeno(123cd)pyrene #	~	0.46	~	~	10631					<0.02	mg/kg	TM4/PM8
Dibenzo(ah)anthracene #	~	0.40	~	~ 🔨	8.61					<0.02	mg/kg	TM4/PM8
Benzo(ghi)perylene #	~	0.55	~	~	1.09					<0.04	mg/kg	TM4/PM8
PAH 16 Total	~	5.35	~	~ nt	13.74					<0.38	mg/kg	TM4/PM8
Coronene	~	0.19	~	OTSC	0.30					<0.02	mg/kg	TM4/PM8
PAH 17 Total	~	5.54	~	<u> </u>	14.00					<0.40	mg/kg	TM4/PM8

#### Jones Environmental Laboratory Client Name: O'Callaghan Moran & Associates

Client Name:	O'Callaghan Moran & A
Reference:	09-188-01
Location:	Tipperary Town Landfill
Contact:	Barry Sexton
JE Job No ·	09/3607

Report :

CEN 10:1 Leachates (expressed as mg/kg)

Solids: V=60g VOC jar, J=250g glass jar, T=plastic tub

JE J00 NO	03/3007									-		
J E Sample No.	1-2	3-4	5-6	7-8	9-10							
Sample ID	TP-1	TP-4	TP-11	TP-12	TP-15							
Depth	-	-	-	-	-							
COC No / misc										Please se	e attached r	notes for all
Containers	JΤ	JΤ	JΤ	JΤ	JΤ					abbrevi	ations and a	acronyms
Sample Date	03/11/09	03/11/09	03/11/09	03/11/09	03/11/09							
Sample Type	Soil	Soil	Soil	Soil	Soil							
Batch Number	1	1	1	1	1						Unite	Method
Date of Receipt	05//11/09	05//11/09	05//11/09	05//11/09	05//11/09					LOD	onita	No.
Arsenic	~	0.08	~	~	0.115					<0.01	mg/kg	TM30
Barium	~	0.48	~	~	1.92					<0.03	mg/kg	TM30
Cadmium	~	<0.01	~	~	<0.01					<0.01	mg/kg	TM30
Chromium	~	<0.02	~	~	<0.02					<0.02	mg/kg	TM30
Copper	~	<0.12	~	~	<0.12					<0.12	mg/kg	TM30
Mercury	~	<0.001	~	~	<0.001					<0.001	mg/kg	TM30
Molybdenum	~	0.13	~	~	0.95					<0.05	mg/kg	TM30
Nickel	~	<0.06	~	~	0.05					<0.06	mg/kg	TM30
Lead	~	<0.1	~	~	<0.1					<0.1	mg/kg	TM30
Antimony	~	0.03	~	~	0.24					<0.03	mg/kg	TM30
Selenium	~	<0.03	~	~	<0.03					<0.03	mg/kg	TM30
Zinc	~	0.05	~	~	0.09			<u>ر</u> ه.		<0.04	mg/kg	TM30
Chloride	~	70	~	~	1847			arus		<1	mg/kg	TM38
Fluoride	~	<1	~	~	<1			sthe		<1	mg/kg	TM38
Sulphate (Soluble)	~	503	~	~	934		17. 201			<1	mg/kg	TM38
Phenol	~	<1	~	~	<1	C.	ottot			<1	mg/kg	TM26
DOC	~	120	~	~	190	0050	ed			<20	mg/kg	TM060
TDS	~	2340	~	~	5860	DITTO				<400	mg/kg	TM20
					Ś	onstr						
					pect	own .						
					in the							
				×	to Aut							
				S	COL							
				ont								
				CONSC .								
				C								

Jones Environme	ntal Lal	boratory	,										
Client Name:	O'Callagi	han Morar	n & Assoc	iates			Report :		Liquids				
Reference:	09-188-0	1											
Location:	Tipperary	/ Town La	Indfill										
Contact:	Barry Se	xton					Liquids/pr	oducts: V=	40ml vial, C	G=glass bott	tle, P=plast	ic bottle	
JE Job No.:	09/3607	•					H=H <sub>2</sub> SO <sub>4</sub> , 2	Z=ZnAc, N=	NaOH, HN	=HN0 <sub>3</sub>			
J E Sample No.	11-16												
Sample ID	SW1												
Depth	-												
COC No / misc	:										Please se	ee attached	notes for all
Containers	VHPG										abbrev	iations and	acronyms
Sample Date	03/11/09												
Sample Type	Water												
Batch Number	1											Unite	Method
Date of Receipt	05/11/09										LOD	Units	No.
pH <sup>#</sup>	7.29										<0.01	pH units	TM19/PM11
Electrical Conductivity#@25°	707										<100	µS/cm	TM28/PM11
Total Oxidised Nitrogen as N	0.50										<0.05	mg/l	TM038W
Ammonia Total as NH3 <sup>#</sup>	1.32										<0.2	mg/l	TM038W
Total Suspended Solids	6										<10	mg/l	TM037W
Dissolved Oxygen	5										<1	ma/l	TM059
BOD settled	3										<1	ma/l	TM058W
COD	23										<7	ma/l	TM057W
005	20												
Areania disselyed #	37										<25	ug/l	TM 030W/
Arsenic - dissolved	34										~12.0	µg/l	TM 030W
Codmium dissolved	-0 F										<0.5	µg/1	TM 020W
Cadmium - dissolved	<0.5							150.			<0.5	μg/ι	TM 030W
Copper - dissolved	~/							net			-1	μg/ι	TM 030W
Mercury - dissolved	<1 10						1. 4	or .			<1	μg/i	TN 030W
NICKEI - dissolved "	<2						oup an,				<2	µg/i	
Lead - dissolved "	<5					يق	2 for				<5	µg/I	TM 030W
Zinc - dissolved "	<3					1Po.	e				<3	µg/I	TM 030W
Iron - dissolved "	182					D Purede					<20	µg/l	TM 030W
Manganese - dissolved *	<2				فني	orner					<2	µg/l	TM 030W
Calcium - dissolved	103.4				SPer	0 <sup>34</sup>					<0.03	mg/l	TM 030W
Magnesium - dissolved	9.42				A HI BU						<0.02	mg/l	TM 030W
				1	C Br								
Sulphate <sup>#</sup>	26.99			2	001						<0.05	mg/l	TM038W
Chloride <sup>#</sup>	28.16			ent							<0.3	mg/l	TM038W
Fluoride	<0.3			COLS							<0.3	mg/l	TM027W
Total Alkalinity as CaCO3 <sup>#</sup>	240			U							<1	mg/l	TM032W
Total Cyanide*	<40										<40	μg/l	subcontracted
Chromium - total	<1.5										<1.5	µg/l	TM 030W
Phosphorous - dissolved #	70										<5	µg/l	TM 030W
Potassium - dissolved	5.44										<0.04	mg/l	TM 030W
Sodium - dissolved	17.54										<0.15	mg/l	TM 030W
	l												
	1	1	1	i i	i i	1	1	1	1	1		1	1

Jones Environme	ntal Lab	boratory	1										
Client Name:	O'Callagh	han Morar	n & Assoc	iates			Report :		Liquids				
Reference:	09-188-0	1											
Location:	Tipperary	/ Town La	ndfill										
Contact:	Barry Sex	xton					Liquids/pr	oducts: V=	-40ml vial, C	G=glass bottl	le, P=plasti	c bottle	
JE Job No.:	09/3607						H=H <sub>2</sub> SO <sub>4</sub> , 2	Z=ZnAc, N=	NaOH, HN	=HN0 <sub>3</sub>	-		
J E Sample No.	11-16												
Sample ID	SW1												
Depth	-												
COC No / misc											Please se	e attached	notes for all
Containers	VHPG										abbrevi	ations and	acronyms
Sample Date	03/11/09												
Sample Type	Water												
Batch Number	1										1.00	11-11-	Method
Date of Receipt	05/11/09										LOD	Units	No.
VOCs <sup>#</sup>	See Tab										<1-27	µg/l	TM15/PM10
SVOCs (dissolved)	See Tab										<10	µg/l	TM16/PM9
Combined OP & OC Pesticides	See Tab										<0.01	µg/l	TM042
PAH 16 (Dissolved) MS													
Naphthalene	<0.1										<0.1	μq/l	TM4/PM9
Acenaphthylene	<0.08										<0.08	µa/l	TM4/PM9
Acenaphthene	<0.1										<0.1	µa/l	TM4/PM9
Fluorene	<0.07										<0.07	ua/l	TM4/PM9
Phenanthrene	<0.07										< 0.07	μg/l	TM4/PM9
Anthracene	<0.08							<b>.</b>			<0.08	μg/l	TM4/PM9
Fluoranthene	<0.09							150			<0.09	μg/l	TM4/PM9
Pyrene	<0.12							ther			<0.12	μg/l	TM4/PM9
Benz(a)anthracene	<0.09						to to	0			<0.09	μg/l	TM4/PM9
Chrysene	<0.1						Officit St				<0.1	μg/l	TM4/PM9
Benzo(bk)fluoranthene	<0.26					050	eg,				<0.26	μg/l	TM4/PM9
Benzo(a)pyrene	<0.12					OTHON	<b>P</b>				<0.12	μg/l	TM4/PM9
Indeno(123cd)pyrene	<0.1					onerre					<0.1	μg/l	TM4/PM9
Dibenzo(ah)anthracene	<0.1				pect	WILL					<0.1	μg/l	TM4/PM9
Benzo(ghi)perylene	<0.12				(install	0					<0.12	μg/l	TM4/PM9
PAH 16 Total	<1.60			4	to Vite						<1.60	μg/l	TM4/PM9
				9	COX.								
EPH (C8-C40) (dissolved) # SS	<10			d'C							<10	µg/ I	TM5/PM9
Mineral Oil (interpretation &	<10			anset							<10	µg/ l	TM5/PM9
aalalatiaa)				00									

#### Client Name: O'Callaghan Moran & Associates **SVOC Report :** LIQUID Reference: 09-188-01 Location: Tipperary Town Landfill Contact: Barry Sexton 09/3607 JE Job No.: J E Sample No 11-16 SW1 Sample II Dept COC No / mis Please see attached notes for all abbreviations and acronyms VHPG Container Sample Date 03/11/09 Sample Type Water Batch Numbe Method LOD Units No. Date of Receip 05/11/09 Phenols TM16/PM9 2-Chlorophenol <10 <10 µg/ I TM16/PM9 2-Methylphenol <10 <10 µg/ I 2-Nitrophenol <10 <10 TM16/PM9 µg/ I 2,4-Dichlorophenol <10 <10 TM16/PM9 µg/ I 2,4-Dimethylphenol <10 <10 µg/ I TM16/PM9 2,4,5-Trichlorophenol <10 <10 µg/ I TM16/PM9 2,4,6-Trichlorophenol <10 TM16/PM9 <10 µg/l 4-Chloro-3-methylphenol <10 <10 µg/ I TM16/PM9 4-Methylphenol <10 <10 TM16/PM9 µg/ I 4-Nitrophenol <10 <10 TM16/PM9 ua/ I Pentachlorophenol <10 <10 µg/ I TM16/PM9 <10 <10 TM16/PM9 Phenol µg/ I PAHs TM16/PM9 2-Chloronaphthalene <10 <10 µg/l other USE. 2-Methylnaphthalene <10 <10 µg/ I TM16/PM9 <10 TM16/PM9 µg/ I Naphthalene only. any <10 TM16/PM9 Acenaphthylene µg/ I <10 µg/ I TM16/PM9 Acenaphthene convision purpose of the second <10 TM16/PM9 Fluorene µg/ I Phenanthrene <10 µg/ I TM16/PM9 <10 TM16/PM9 Anthracene µg/ I SEE <10 TM16/PM9 µg/l Fluoranthene PAH TM16/PM9 Pyrene <10 µg/ I RESULTS <10 TM16/PM9 Benz(a)anthracene µg/ I <10 TM16/PM9 Chrvsene µg/ I Benzo(bk)fluoranthene <10 µg/ I TM16/PM9 <10 µg/ I TM16/PM9 Benzo(a)pyrene TM16/PM9 <10 µg/l Indeno(123cd)pyrene Consent TM16/PM9 Dibenzo(ah)anthracene <10 µg/ I <10 TM16/PM9 Benzo(ghi)perylene µg/ I Phthalates Bis(2-ethylhexyl) phthalate <10 <10 µg/ I TM16/PM9 Butylbenzyl phthalate <10 <10 TM16/PM9 µg/ I TM16/PM9 Di-n-butyl phthalate <10 <10 µg/ I Di-n-Octyl phthalate <10 <10 µg/ I TM16/PM9 Diethyl phthalate <10 <10 TM16/PM9 µg/ I Dimethyl phthalate <10 TM16/PM9 <10 µg/ I Other SVOCs 1,2-Dichlorobenzene <10 <10 TM16/PM9 µg/ I 1,2,4-Trichlorobenzene <10 TM16/PM9 <10 µg/l 1 3-Dichlorobenzene <10 <10 µg/ I TM16/PM9 1,4-Dichlorobenzene <10 TM16/PM9 <10 µg/ I 2-Nitroaniline TM16/PM9 <10 <10 µg/ I 2 4-Dinitrotoluene TM16/PM9 <10 <10 µg/ I 2,6-Dinitrotoluene <10 <10 TM16/PM9 µg/ I 3-Nitroaniline <10 <10 TM16/PM9 µg/ I TM16/PM9 4-Bromophenylphenylether <10 <10 µg/ I 4-Chloroaniline <10 <10 TM16/PM9 µg/ I 4-Chlorophenylphenylether <10 TM16/PM9 <10 µg/l TM16/PM9 4-Nitroaniline <10 <10 µg/ I Azobenzene <10 <10 TM16/PM9 µg/ I Bis(2-chloroethoxy)methane <10 <10 TM16/PM9 µg/ I Bis(2-chloroethyl)ether <10 <10 µg/ I TM16/PM9 Carbazole <10 <10 µg/ I TM16/PM9 <10 TM16/PM9 Dibenzofuran <10 µg/ I Hexachlorobenzene <10 <10 µg/ I TM16/PM9 Hexachlorobutadiene <10 <10 TM16/PM9 µg/ I <10 <10 TM16/PM9 Hexachlorocyclopentadiene ua/ I Hexachloroethane <10 <10 µg/ I TM16/PM9 Isophorone <10 <10 µg/ I TM16/PM9 <10 <10 TM16/PM9 N-nitrosodi-n-propylamine µg/ I

Please include all sections of this report if it is reproduced

Nitrobenzene

<10

Jones Environmental Laboratory

TM16/PM9

<10

µg/ I

#### Client Name: O'Callaghan Moran & Associates **VOC Report :** LIQUID Reference: 09-188-01 Location: Tipperary Town Landfill Contact: Barry Sexton 09/3607 JE Job No.: J E Sample No 11-16 SW1 Sample II Dept COC No / mis VHPG Container Please see attached notes for all abbreviations and acronyms Sample Date 03/11/09 Sample Type Water Batch Numbe Method LOD Units No. Date of Receip 05/11/09 Dichlorodifluoromethane TM15/PM10 <2 <2 µg/l TM15/PM10 Methyl Tertiary Butyl Ether <2 <2 µg/l <3 <3 TM15/PM10 Chloromethane µg/l TM15/PM10 Vinvl Chloride <2 <2 µg/l <1 TM15/PM10 Bromomethane <1 µg/l <3 TM15/PM10 <3 Chloroethane # µg/l Trichlorofluoromethane # <3 <3 µg/l TM15/PM10 <3 <3 TM15/PM10 1 1-Dichloroethene # μg/l NA TM15/PM10 Carbon Disulphide # <3 μg/l Dichloromethane <sup>#</sup> <3 <3 µg/l TM15/PM10 <3 <3 TM15/PM10 trans-1-2-Dichloroethene # µg/l <3 TM15/PM10 1,1-Dichloroethane <3 µg/l <3 <3 TM15/PM10 cis-1-2-Dichloroethene# µg/l TM15/PM10 2.2-Dichloropropane <1 <1 µg/l other use. Bromochloromethane # <2 <2 µg/l TM15/PM10 TM15/PM10 <3 <3 Chloroform # µg/l only. any <3 TM15/PM10 1,1,1-Trichloroethane # <3 μg/l 1,1-Dichloropropene # <3 <3 µg/l TM15/PM10 conviet on purpose <2 <2 TM15/PM10 Carbon tetrachloride # μg/l 1,2-Dichloroethane # <2 <2 µg/l TM15/PM1 <3 TM15/PM10 Benzene # <3 µg/l TM15/PM10 Trichloroethene <3 <3 µg/l 1,2-Dichloropropane # <2 <2 TM15/PM10 µg/l <3 <3 TM15/PM10 Dibromomethane <sup>#</sup> µg/l Bromodichloromethane# <3 <3 TM15/PM10 µg/l cis-1-3-Dichloropropene\* <2 <2 µg/l TM15/PM10 <3 TM15/PM10 Toluene # <3 μg/l Consent trans-1-3-Dichloropropene\* <2 <2 µg/l TM15/PM1 <2 <2 TM15/PM10 1,1,2-Trichloroethane µg/l Tetrachloroethene # <3 <3 µg/l TM15/PM10 <2 <2 TM15/PM10 1,3-Dichloropropane # µg/l Dibromochloromethane<sup>#</sup> <2 <2 µg/l TM15/PM10 <2 <2 TM15/PM10 1.2-Dibromoethane µg/l Chlorobenzene <sup>#</sup> <2 <2 µg/l TM15/PM10 <2 <2 TM15/PM10 1.1.1.2-Tetrachloroethane # µg/l Ethylbenzene # <3 <3 µg/l TM15/PM1 <5 TM15/PM10 p/m-Xylene<sup>#</sup> <5 µg/l o-Xylene <sup>#</sup> <3 <3 µg/l TM15/PM10 <2 Stvrene <2 TM15/PM10 µg/l Bromoform # <2 <2 µg/l TM15/PM10 Isopropylbenzene # <3 <3 TM15/PM10 µg/l 1,1,2,2-Tetrachloroethane <4 <4 TM15/PM10 μg/l <2 <2 TM15/PM10 Bromobenzene µg/l 1,2,3-Trichloropropane# <3 <3 µg/l TM15/PM1 <3 <3 TM15/PM10 Propylbenzene<sup>4</sup> µg/l 2-Chlorotoluene # <3 <3 TM15/PM10 µg/l 1,3,5-Trimethylbenzene# <3 <3 TM15/PM10 µg/l 4-Chlorotoluene # <3 <3 µg/l TM15/PM10 <3 <3 TM15/PM10 tert-Butylbenzene\* µg/l 1,2,4-Trimethylbenzene# <3 <3 μg/l TM15/PM10 <3 <3 TM15/PM10 sec-Butylbenzene µg/l 4-Isopropyltoluene \* <3 <3 TM15/PM10 μg/l <3 <3 TM15/PM10 1,3-Dichlorobenzene\* µg/l 1,4-Dichlorobenzene# <3 <3 TM15/PM10 µg/l <3 <3 TM15/PM10 n-Butvlbenzene ua/l 1,2-Dichlorobenzene\* <3 <3 TM15/PM10 µg/l 1,2-Dibromo-3-chloropropane <2 <2 TM15/PM10 µg/l 1,2,4-Trichlorobenzene <3 <3 µg/l TM15/PM10 <3 <3 TM15/PM10 Hexachlorobutadiene µg/l

1,2,3-Trichlorobenzene

Naphthalene

<2

<3

Jones Environmental Laboratory

All solid results are expressed on a dry weight basis unless stated otherwise.

8 of 10

TM15/PM1

TM15/PM10

<2

<3

µg/l

µg/l

### Jones Environmental Laboratory

Client Name:	O'Callagi	han Morar	n & Assoc	iates							
Reference:	09-188-0	1									
Location:	Tipperary	/ Town La	ndfill								
Contact:	Barry Se	xton									
JE Job No ·	09/3607										
J E Sample No	11-16		1						1		
Sample ID	SW1										
Denth	-										
200											
Containers	VHPG										
Sample Date	03/11/09										
Sample Type	Water										
Batch Number	1										Mothod
Date of Receipt	05/11/09								LOD	Units	Number
Combined Posticide Suite	03/11/03										
	<0.01								<0.01	ug/l	TM042
Mevinnhos	<0.01								<0.01	µg/l	TM042
Alpha BHC	<0.01								<0.01	µg/l	TM042
Reta BHC	<0.01								<0.01	µg/i	TM042
	<0.01								<0.01	µg/i	TM042
Gamilla-DHC Diazinon	<0.01								<0.01	µg/i	TM042
Mothyl Parathian	<0.01								<0.01	µg/i	TM042
Ethyl Parathion (Parathion)	<0.01								<0.01	µg/i	TM042
Ethyl Parathion (Parathion)	<0.01								<0.01	µg/i	TM042
Fepitachion	<0.01								<0.01	µg/i	TM042
Aldrin	<0.01								<0.01	µg/i	TM042
Malathion	<0.01								<0.01	µg/i	TM042
Hantachlar Enovida	<0.01								<0.01	µg/i	TM042
	<0.01								<0.01	µg/l	TM042
Dieldrin	<0.01								<0.01	µg/l	TM042
	<0.01								<0.01	µg/l	TM042
Fndosulfan II	<0.01							01.	<0.01	µg/i	TM042
	<0.01							, 150	<0.01	µg/l	TM042
Fthion	<0.01							net	<0.01	µg/l	TM042
Endrin	<0.01						,C	N.	<0.01	µg/l	TM042
Endosulfan Sulphate	<0.01						19. 20		<0.01	µg/l	TM042
	<0.01					5	101		<0.01	µg/l	TM042
Methoxychlor	<0.01					-050.0	8,		<0.01	µg/l	TM042
Azinphos Methyl	<0.01					all all	ſ		<0.01	µ=3.1	TM042
Disulfoton	<0.01					Prov			< 0.01	µg/l	TM042
					di	not .				15	
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# NOTES TO ACCOMPANY ALL SCHEDULES AND REPORTS

### SOILS

Please note we are only MCERTS accredited for sand, loam and clay and any other matrix is outside our scope of accreditation.

Where an MCERTS report has been requested, you will be notified within 48 hours of any samples that have been identified as being outside our MCERTS scope. As validation has been performed on clay, sand and loam, only samples that are predominantly these matrices, or combinations of them will be within our MCERTS scope. Your final report will reflect this, with non-MCERTS results on separate pages.

It is assumed that you have taken representative samples on site and require analysis on a representative subsample. Stones will generally be included unless we are requested to remove them.

All samples will be discarded one month after the date of reporting, unless we are instructed to the contrary. If we are instructed to keep samples, a storage charge of £1 (1.5 Euros) per sample per month will be applied until we are asked to dispose of them.

If you have not already done so, please send us a purchase order if this is required by your company.

Where appropriate please make sure that our detection limits are suitable for your needs, if they are not, please notify us immediately.

All analysis is carried out on as received samples and reported on a dry weight basis unless stated otherwise. Results are not surrogate corrected.

Asbestos screens where requested will be undertaken by a UKAS accredited laboratory.

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### WATERS

otheruse Please note we are not a Drinking Water Inspectorate (DWI) Approved Laboratory . It is important that detection limits are carefully considered when requesting water analysis.

UKAS accreditation applies to tap water, surface water and groundwater only, any other liquids are outside ion our scope of accreditation

As surface waters require different sample preparation to groundwaters the laboratory must be informed of the water type when submitting samples. All samples are treated as groundwaters and anlalysis performed on settled samples unless we are instructed otherwise.

### **DEVIATING SAMPLES**

Samples must be received in a condition appropriate to the requested analyses. All samples should be submitted to the laboratory in suitable containers with sufficient ice packs to sustain an appropriate temperature for the requested analysis. If this is not the case you will be informed and any analysis that may be compromised highlighted on your schedule/ report by the use of a symbol.

The use of any of the following symbols indicates that the sample was deviating and the test result may be unreliable:

- \$ sample temperature on receipt considered inappropriate for analysis requested
- ۸ samples exceeding recomended holding times
- samples received in inappropriate containers (e.g. volatile samples not submitted in VOC jars/vials) &
- no sampling date given, unable to confirm if samples are with acceptable holding times

#### ABBREVIATIONS and ACRONYMS USED

- # UKAS accredited
- M MCERTS accredited
- NAD No Asbestos Detected
- ND None Detected (usually refers to VOC and/SVOC TICs)
- SS Calibrated against a single substance
- \* analysis subcontracted to a Jones Environmental approved laboratory.
- W Results expressed on as received basis
- + Failed AQC results should be considered as indicative only and are not accredited.
- ++ Result outside calibration range, may be possible to re-run with higher detection limits

STORY ST	I STORY	Single       And       And <t< th=""><th></th><th></th><th></th><th>The town</th></t<>				The town
Ф. Полие, Пол. О. Полиента, Поли Полиента, Поли Полиента, Поли Поли.         Вани Полиента, Поли Поли Поли.         Вани Поли Поли         Вани Поли Поли.         Вани Пол	Monte         Monte <th< td=""><td>And         Полован         Алилован         Полован         Полован</td><td>USTODY UCCLOUN HORN</td><td>t Allocater,</td><td>SAMPLER: B. Sorbar</td><td>Jones</td></th<>	And         Полован         Алилован         Полован	USTODY UCCLOUN HORN	t Allocater,	SAMPLER: B. Sorbar	Jones
	Поли Поли Поли Поли Поли         Поли Поли Поли Поли Поли         Поли Поли Поли Поли Поли           255 - 2.4 (17.4 - 10.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)         100.1 (17.4 m)           201 1 2 - 2.4	Intel Direction         Interesting	and ritual, al	ward Xirel, Com.	MOBILE: EMAIL REPORT TO: R. Sox for a	Environmental
Set 2 - 次月 15 年4 (2)     Control 1 - Control 1	Sist 3 - 2 - 51 - 51 - 51 - 51 - 51 - 51 - 51	Sint 3	(PM): Bany Sonta		cc REPORT TO:	
Control 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	ПСП-ПСИ-ПСИ-ПСИ-ПСИ-ПСИ-ПСИ-ПСИ-ПСИ-ПСИ-	M. M. Town         M. M. Start         M. M. Start         M. M. Start         M. Start <thm. start<="" th="">         M. Start         <thm. start<="" td="" th<=""><td>0353 - 846156</td><td>040</td><td>INVOICE TO: (if different to report)</td><td></td></thm.></thm.>	0353 - 846156	040	INVOICE TO: (if different to report)	
Control Contro Control Control Control Control Control Control Control Control	Общест         Полон         <		CH-188-01		QUOTE NUMBER: P.O No:	Chain of Custody sheet page of
Rest     Image: Construction of the con		Image: Net Control         Control         Image: Con	orcary Town L	Land Hill	ANALYSIS REQUIRED including SUITE names	COMMENTS
10.1 「010 」 「100 」 □10 」 □10 」 □10 」 □10 □10	100/100         000/100	100/// Image: Control C	se tick	FOR LABORATORY USE ONLY		Notes: e.g. Heavily contaminated samples
Sume are free on the first and	Junctification         Junctification         Junctification         Junctification           Subtract in control in contr	Subtract Finder Stress       Market Finder Stress       Market Finder Stress         Subtract Finder Stress       June       June       Market Finder Stress         Subtract Finder Stress       June       June       June       Market Finder Stress         Subtract Finder Stress       June       June       June       June       June         Super Stress       June	4 DAY Other	AVERAGE COOL BOX TEMP (if required):	Le Contra and and and and and and and and and an	e.g. "High PAHs expected".
			3 DAY     SAMPI F INFORMATION (note: 5	SAMPLE RECIEPT CONDITION: S = Soit W=Water P=Product(oit)	かって いってい いっちん いっしん ししん	NOTE: If an MCERTS report is
			Sample ID	Matrix Date Time Depth in Metres on	Proventing and	required into must be requested when samples are scheduled.
13     25     14     15     14     15       13     25     14     15     14     15     14       13     25     14     15     14     15     14       14     15     16     14     15     14     15       15     15     16     14     14     14     15       16     16     16     16     14     14     15       17     16     16     16     16     16     16       18     16     16     16     16     16     16       19     16     16     16     16     16     16	H     H     H     H     H     H       1     2     0     0     0     0     0       1     2     0     0     0     0     0     0       1     2     0     0     0     0     0     0       1     2     0     0     0     0     0     0     0       1     2     0     0     0     0     0     0     0     0       1     1     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0     0       1     0     0     0     0     0     0     0     0     0 <td>1     1<td></td><td>V</td><td></td><td></td></td>	1     1 <td></td> <td>V</td> <td></td> <td></td>		V		
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Date: Name: Date: Consignment note No: Time: Of Courier Company:	Date:     Date:     Uate:     Uote:     Uote:     Uote:       ions including known hazards (eg suspected asbestos)     0f:     1me:     Courier Company:	Date:     Date:     Uate:     Uate:     Consignment note No:       ions including known hazards (eg suspected asbestos)     Time:     Counter Company:	0		RECEIVED BY:	METHOD of SHIPMENT
	tions including known hazards (eg suspected abestos)	tions including known hazards (eg suspected asbestos)	and have	Date: Time.	Name: A Date: Of Time:	Consignment note No: Courrier Company:

Jones Environmental Laboratory Unit 3 Deeside Point, Zone 3 Deeside Industrial Park, Deeside, Flintshire CH5 2UA Tel: 01244 833 780

JEL 6863

QF-PM1.1 v2

EPA Export 30-11-2011:03:45:43

# Jones Environmental Laboratory



Unit 3 Deeside Point Zone 3 Deeside Industrial Park Deeside CH5 2UA

### Tel: +44 (0) 1244 833780 Fax: +44 (0) 1244 833781



Eight samples were received for analysis on 25th November 2009 which was completed on 8th December 2009. Please find attached our Test Report which should be read with notes at the end of the report and should include all sections if reproduced. All interpretations and opinions are outside the scope of any accreditation, and all results relate only to samples supplied. All analysis is carried out on as received samples and reported on a dry weight basis unless stated otherwise. Results are not surrogate corrected.

J W Farrell- Jones CChem FRSC Chartered Chemist

### Jones Environmental Laboratory

Client Name:	
Reference:	

Location:

O'Callaghan Moran & Associates 09-188-01 Tipperary Town Landfill Report :

Liquids

Liquids/products: V=40ml vial, G=glass bottle, P=plastic bottle

Contact: Barry Sexton

JE Job No.:	09/3874						H=H <sub>2</sub> SO <sub>4</sub> , 2	Z=ZnAc, N=	NaOH, HN	=HN0 <sub>3</sub>			
J E Sample No.	1-6	7-12	13-18	19-24	25-30	31-36	37-42	43-48					
Sample ID	MW1	MW2	MW3	MW4	MW5	MW6	MW7	MW8			1		
Denth	-	-	-	-	-	_	_	-					
COC No / misc											Please se	ee attached	notes for all
Containers	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG			abbiev		acronyms
Sample Date	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09					
Sample Type	Water	Water	Water	Water	Water	Water	Water	Water					
Batch Number	1	1	1	1	1	1	1	1					Mathead
Date of Peccipt	25/11/00	25/11/00	25/11/00	25/11/00	25/11/00	25/11/00	25/11/00	25/11/00			LOD	Units	Nethod No.
#	23/11/09	23/11/09	20/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09			-0.01	nH unita	TM10/DM11
pH Electrical Conductivity <sup>#</sup> @25%	3710	4370	6370	1232	~	1389	~	1490			<100	uS/cm	TM28/PM11
Total Oxidised Nitrogen as N	<0.05	<0.05	<0.05	1 22	~	42 27	~	<0.05			<0.05	ma/l	TM038W
Ammonia Total as NH3 <sup>#</sup>	70.5	43.5	18.1	0.7	~	0.8	~	0.2			<0.2	mg/l	TM038W
Total Dissolved Solids	~	~	~	472	~	947	~	919			<35	mg/l	TM020W
ТОС	~	~	~	5	~	10	~	6			<2	mg/l	TM060W
BOD settled	20	26	9	~	~	~	~	~			<1	mg/l	TM058W
COD	114	183	52	~	~	~	~	~			<7	mg/l	TM057W
								e.					
Arsenic - dissolved #	19.2	17.1	10.3	6	~	6.6	~	6.6			<2.5	µg/l	TM 030W
Boron - dissolved	945	1917	733	25	~	258	~	20			<12	µg/l	TM 030W
Cadmium - dissolved #	<0.5	<0.5	<0.5	<0.5	~	<0.5	only and	<0.5			<0.5	µg/l	TM 030W
Copper - dissolved #	<7	<7	<7	<7	~	<7	-d for	12			<7	µg/l	TM 030W
Mercury - dissolved #	<1	<1	<1	<1	~		~	<1			<1	µg/l	TM 030W
Nickel - dissolved "	4	15	<2	<2 E	~	on P2 rout	~	4			<2	µg/I	TM 030W
Zinc, dissolved <sup>#</sup>	4	5 11	4	5 <3	~ pecto	WILE3	~	0 10			<3	µg/i	TM 030W
Iron - dissolved #	81	52	<20	<20	inst th	<20	~	<20			<20	ug/l	TM 030W
Manganese - dissolved #	903	385	706	116	to vite	342	~	538			<2	µg/l	TM 030W
Calcium - dissolved	122.30	47.91	166.40	119.10	COX-	144.4	~	147.7			<0.03	mg/l	TM 030W
Magnesium - dissolved	42.28	28.96	58.08	9.300	~	14.82	~	19.03			<0.02	mg/l	TM 030W
				COLSC									
Sulphate <sup>#</sup>	6.79	100.53	3.15	14.78	~	104.22	~	11.22			<0.05	mg/l	TM038W
Chloride <sup>#</sup>	235.2	948.6	1703.7	57.9	~	135.9	~	276.2			<0.3	mg/l	TM038W
Fluoride	<0.3	0.3	0.5	<0.3	~	<0.3	~	<0.3			<0.3	mg/l	TM027W
Total Alkalinity as CaCO3 <sup>#</sup>	~	~	~	308	~	388	~	368			<1	mg/l	TM032W
l otal Cyanide*	<40	<40	<40	<40	~	<40	~	<40			<40	μg/l	subcontracted
Phoenborous discoluted #	56	2.0 336	21	10	~	12	~	×1.5 11			<1.0 <5	µg/i	TM 030W
Potassium - dissolved	74.02	127.00	65.60	1.58	~	5.64	~	1.21			<0.04	ma/l	TM 030W
Sodium - dissolved	100.30	352.50	586.30	40.11	~	101.30	~	81.15			<0.15	mg/l	TM 030W
EPH (C8-C40) (dissolved) # SS	352	92	<10	<10	~	<10	~	<10			<10	µg/ I	TM5/PM9
Mineral Oil (interpretation &	<10	<10	<10	<10	~	<10	~	<10			<10	ua/ I	TM5/PM9
calculation)												-9.1	

#### Jones Environmental Laboratory Client Name: O'Callaghan Moran & Associates

Client Name:	O'Callaghan Moran & A
Reference:	09-188-01
Location:	Tipperary Town Landfill
Contact:	Barry Sexton
JE Job No.:	09/3874

Report :

Liquids

 $\label{eq:liquids/products: V=40ml vial, G=glass bottle, P=plastic bottle H=H_2SO_4, Z=ZnAc, N=NaOH, HN=HNO_3$ 

J E Sample No.	1-6	7-12	13-18	19-24	25-30	31-36	37-42	43-48				
Sample ID	MW1	MW2	MW3	MW4	MW5	MW6	MW7	MW8				
Depth	-	-	-	-	-	-	-	-				
COC No / misc										Please se	e attached	notes for all
Containers	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG		abbrev	iations and a	acronyms
Sample Date	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09				
Sample Type	Water	Water	Water	Water	Water	Water	Water	Water				
Batch Number	1	1	1	1	1	1	1	1			Unite	Method
Date of Receipt	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09		LOD	Units	No.
VOCs #	See tab	See tab	See tab	See tab	~	See tab	~	See tab		<1-27	µg/l	TM15/PM10
SVOCs (dissolved)	See tab	See tab	See tab	See tab	~	See tab	~	See tab		<10	µg/l	TM16/PM9
Combined OP & OC Pesticides	See tab	See tab	See tab	See tab	~	See tab	~	See tab		<0.01	µg/l	TM042
PAH 16 (Dissolved) MS												
Naphthalene	42.5	<0.1	<0.1	<0.1	~	<0.1	~	<0.1		<0.1	μg/l	TM4/PM9
Acenaphthylene	<0.08	<0.08	<0.08	<0.08	~	<0.08	~	<0.08		<0.08	μg/l	TM4/PM9
Acenaphthene	1.4	<0.1	<0.1	<0.1	~	<0.1	~	<0.1		<0.1	μg/l	TM4/PM9
Fluorene	0.90	<0.07	<0.07	<0.07	~	<0.07	~	<0.07		<0.07	μg/l	TM4/PM9
Phenanthrene	0.80	<0.07	<0.07	<0.07	~	<0.07	~	<0.07		<0.07	μg/l	TM4/PM9
Anthracene	<0.08	<0.08	<0.08	<0.08	~	<0.08	~	<0.08		<0.08	μg/l	TM4/PM9
Fluoranthene	<0.09	<0.09	<0.09	<0.09	~	<0.09	~	\$0.09		<0.09	μg/l	TM4/PM9
Pyrene	<0.12	<0.12	<0.12	<0.12	~	<0.12	~	×0.12		<0.12	μg/l	TM4/PM9
Benz(a)anthracene	<0.09	<0.09	<0.09	<0.09	~	<0.09	aly and	<0.09		<0.09	μg/l	TM4/PM9
Chrysene	<0.1	<0.1	<0.1	<0.1	~	<0.1	ortor	<0.1		<0.1	μg/l	TM4/PM9
Benzo(bk)fluoranthene	<0.26	<0.26	<0.26	<0.26	~	<0.26	eo ~	<0.26		<0.26	μg/l	TM4/PM9
Benzo(a)pyrene	<0.12	<0.12	<0.12	<0.12	~	0.120	~	<0.12		<0.12	μg/l	TM4/PM9
Indeno(123cd)pyrene	<0.1	<0.1	<0.1	<0.1	~	01 50.1	~	<0.1		<0.1	μg/l	TM4/PM9
Dibenzo(ah)anthracene	<0.1	<0.1	<0.1	<0.1	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	<0.1	~	<0.1		<0.1	μg/l	TM4/PM9
Benzo(ghi)perylene	<0.12	<0.12	<0.12	<0.12	A HIGH	<0.12	~	<0.12		<0.12	μg/l	TM4/PM9
PAH 16 Total	45.60	<1.60	<1.60	<1.60	C St.	<1.60	~	<1.60		<1.60	μg/l	TM4/PM9
				S								
				ent								
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# Jones Environmental Laboratory

	THE LICE														
Client Name:	O'Callaghan Moran & Associates							ort :	Liquids						
Reference:	09-188-01														
Location:	Tipperary Town Landfill														
Contact:	Barry Sexton							Liquids/products: V=40ml vial, G=glass bottle, P=plastic bottle							
JE Job No.:	09/3874			H=H <sub>2</sub> SO <sub>4</sub> , Z=ZnAc, N=NaOH, HN=HN0 <sub>3</sub>											
J E Sample No.	1-6	7-12	13-18	19-24	31-36	43-48									
Sample ID	MW1	MW2	MW3	MW4	MW6	MW8									
Depth	-	-	-	-	-	-									
COC No / misc											Please se	e attached	notes for all		
Containers	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG					abbrev	iations and a	acronyms		
Sample Date	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09									
Sample Type	Water	Water	Water	Water	Water	Water									
Batch Number	1	1	1	1	1	1							Method		

Sample Type	vvaler	water	water	water	water	water			 		1	
Batch Number	1	1	1	1	1	1				LOD	Units	Method
Date of Receipt	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09						No.
Dichlorodifluoromethane	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Methyl Tertiary Butyl Ether	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Chloromethane #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
Vinyl Chloride	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Bromomethane	<1	<1	<1	<1	<1	<1				<1	µg/l	TM15/PM10
Chloroethane #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
Trichlorofluoromethane #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1,1-Dichloroethene #	<6	<6	<6	<6	<6	<6				<6	µg/l	TM15/PM10
Carbon Disulphide <sup>#</sup>	NA	NA	NA	NA	NA	NA				<3	µg/l	TM15/PM10
Dichloromethane #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
trans-1-2-Dichloroethene#	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1.1-Dichloroethane #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
cis-1-2-Dichloroethene#	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
2,2-Dichloropropane	<1	<1	<1	<1	<1	<1				<1	µg/l	TM15/PM10
Bromochloromethane #	<2	<2	<2	<2	<2	<2				<2	ug/l	TM15/PM10
Chloroform #	<3	<3	<3	<3	<3	<3				<3	ug/l	TM15/PM10
1 1 1-Trichloroethane #	<3	<3	<3	<3	<3	<3		150.		<3	ug/l	TM15/PM10
1 1-Dichloropropene <sup>#</sup>	<3	<3	<3	<3	<3	<3		not		<3	ug/l	TM15/PM10
Carbon tetrachloride #	<2	<2	<2	<2	<2	<2		SU.		<2	µg/l	TM15/PM10
1 2 Dichloroethane #	<2	<2	<2	<2	<2	<2	and and			<2	µg/l	TM15/PM10
Ronzono #	-2	-2	-2	-2	-2	-2	Or of			-2	µg/1	TM15/PM10
Delizelle	~3	-0	-0	~3	-0	-2050	92			~3	µg/i	TM15/PM10
	< 3	<3	10	< 2	< 2	JI S	50			< 2	µg/i	TM15/PM10
1,2-Dichloropropane	< <u>2</u>	< <u>2</u>	< <u>2</u>	< <u>2</u>	< <u>2</u>	Plazedt				< <u>2</u>	µg/i	TM15/PM10
Dibromomethane"	<3	<3	<3	<3	<3	Orofa				<3	µg/i	TM15/PM10
Bromodichloromethane"	<3	<3	<3	<3	<3 0	AV1 <3				<3	µg/I	TM15/PM10
cis-1-3-Dichloropropene"	<2	<2	<2	<2	- in h	∪ < <u>2</u>				<2	µg/I	TM15/PM10
Toluene "	<3	<3	<3	<3	01 300	<3				<3	µg/l	TM15/PM10
trans-1-3-Dichloropropene*	<2	<2	<2	<2	08 <sup>2</sup>	<2				<2	µg/l	TM15/PM10
1,1,2-Trichloroethane #	<2	<2	<2	<2	° <2	<2				<2	µg/l	TM15/PM10
Tetrachloroethene #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1,3-Dichloropropane #	<2	<2	<2	Nº T	<2	<2				<2	µg/l	TM15/PM10
Dibromochloromethane#	<2	<2	<2	C <sup>022</sup>	<2	<2				<2	µg/l	TM15/PM10
1,2-Dibromoethane #	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Chlorobenzene <sup>#</sup>	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
1,1,1,2-Tetrachloroethane #	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Ethylbenzene #	4	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
p/m-Xylene <sup>#</sup>	9	<5	<5	<5	<5	<5				<5	µg/l	TM15/PM10
o-Xylene <sup>#</sup>	5	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
Styrene #	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Bromoform #	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
Isopropylbenzene #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1,1,2,2-Tetrachloroethane	<4	<4	<4	<4	<4	<4				<4	µg/l	TM15/PM10
Bromobenzene #	<2	<2	<2	<2	<2	<2				<2	µg/l	TM15/PM10
1,2,3-Trichloropropane#	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
Propylbenzene #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
2-Chlorotoluene #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1,3,5-Trimethylbenzene#	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
4-Chlorotoluene #	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
tert-Butvlbenzene#	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1.2.4-Trimethylbenzene <sup>#</sup>	8	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
sec-Butylbenzene #	<3	<3	<3	<3	<3	<3				<3	ug/l	TM15/PM10
4-Isopropyltoluene #	10	<3	<3	<3	<3	<3				<3	µa/l	TM15/PM10
1.3-Dichlorobenzene <sup>#</sup>	<3	<3	<3	<3	<3	<3				<3	µg/l	TM15/PM10
1 4-Dichlorobenzene <sup>#</sup>	<3	<3	<3	<3	<3	<3				<3	ug/l	TM15/PM10
n Butylbonzono <sup>#</sup>	-3	-3	-3	-3	-3	-3				-0	µg/1	TM15/PM10
	-3	~3	~3	-3	-3	-0				-3	µg/i	TM15/DM10
1,2-Dicritoropenzene	~0	~>	~0	-0	-0	~0				-0	µg/i	TM15/DM440
1.2.4 Trichlorchonzono	~2	~2	~2	~2	~2	~2				~2	μg/ι	TM15/PW10
1,2,4-11101000enzene	< 3 - 2	< S - C	< 3 - 2	< 3 - 2	< 3 - 2	< 3 - 2				< 3 - 2	µg/i	TM15/PM10
Hexachlorobutadiene "	< 3	< 3	< 3	< 3	< 3	< 3				< 3	µg/I	TM45/PM10
Naphthalene	68	<2	<2	<2	<2	<2				<2	µg/i	TM15/PM10
1,2,3-Trichlorobenzene	<3	<3	<3	<3	<3	<3				<3	µg/l	1M15/PM10

### Jones Environmental Laboratory Client Name: O'Callaghan Moran & Associates

09/3874

Client Name Reference: Location: Contact:

JE Job No.:

O'Callaghan Moran & Asso 09-188-01 Tipperary Town Landfill Barry Sexton

#### SVOC Report : Liquids

 $\label{eq:Liquids/products: V=40ml vial, G=glass bottle, P=plastic bottle H=H_2SO_4, Z=ZnAc, N=NaOH, HN=HN0_3$ 

J E Sample No.	1-6	7-12	13-18	19-24	31-36	43-48						
Sample ID	MW1	MW2	MW3	MW4	MW6	MW8						
Depth	-	-	-	-	-	-						
COC No / misc										Please se	e attached	notes for all
Containers	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG				abbrev	iations and a	acronyms
Somela Data	22/11/00	22/11/00	22/11/00	22/11/00	22/11/00	22/11/00						
Sample Date	23/11/09	23/11/08	23/11/09	23/11/09	23/11/03	23/11/09						
Sample Type	Water	vvalei	Water	Water	vvalei	Water						Marthaut
Batch Number	1	1	1	1	1	1				LOD	Units	Method
Date of Receipt	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	1					NO.
Phenols												
2-Chlorophenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
2-Methylphenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
2-Nitrophenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
2,4-Dichlorophenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
2,4-Dimethylphenol	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
2,4,5-Trichlorophenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
2,4,6-Trichlorophenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
4-Chloro-3-methylphenol	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
4-Methylphenol	<10	<10	<10	<10	<10	<10				<10	ua/ I	TM16/PM9
4-Nitrophenol	<10	<10	<10	<10	<10	<10				<10	ug/ 1	TM16/PM9
Pentachloronhenol	<10	<10	<10	<10	<10	<10				<10	ug/1	TM16/PM9
Phenol	<10	<10	<10	<10	<10	<10				<10	P9/1	TM16/DM0
DALLa	10	10	10	10	10	10				10	µg/ i	TIVITO/FIVI3
PARS	- 10			-10						-10	116/1	TMAC/DMAC
	<10	<10	<10	<10	<10	<10				<10	μg/ I	1 M16/PM9
2-Methylnaphthalene	19	<10	<10	<10	<10	<10	1			<10	µg/ I	IM16/PM9
Naphthalene								.0.		<10	µg/ I	TM16/PM9
Acenaphthylene								15		<10	μg/ I	TM16/PM9
Acenaphthene								net		<10	µg/ I	TM16/PM9
Fluorene								0,1		<10	µg/ I	TM16/PM9
Phenanthrene							13. 20			<10	µg/ I	TM16/PM9
Anthracene						_	orson			<10	µg/ I	TM16/PM9
Fluoranthene						60	60			<10	µg/ I	TM16/PM9
Pvrene			SEE PAH	RESULTS		18.1	t.			<10	ua/ I	TM16/PM9
Benz(a)anthracene						No. Con				<10	ug/ 1	TM16/PM9
					×	orstr				<10	ug/1	TM16/PM0
					Sec.	WIL				<10	µg/1	TM16/DM0
Benzo(bk)fluorantnene					inst it	0				<10	μg/ 1	TM10/FIVI9
Benzo(a)pyrene					of the					<10	µg/ 1	TIVITO/PIVI9
Indeno(123cd)pyrene					5.23					<10	µg/ I	TM16/PM9
Dibenzo(ah)anthracene				S						<10	µg/ I	TM16/PM9
Benzo(ghi)perylene				<u> </u>	)*					<10	µg/ I	TM16/PM9
Phthalates				Selt								
Bis(2-ethylhexyl) phthalate	<10	<10	<10	C 0810	<10	<10				<10	µg/ I	TM16/PM9
Butylbenzyl phthalate	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
Di-n-butyl phthalate	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
Di-n-Octyl phthalate	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
Diethyl phthalate	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
Dimethyl phthalate	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
Other SVOCs												
1,2-Dichlorobenzene	<10	<10	<10	<10	<10	<10				<10	μα/Ι	TM16/PM9
1 2 4-Trichlorobenzene	<10	<10	<10	<10	<10	<10				<10	ug/1	TM16/PM9
1 3-Dichlorobenzene	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM0
1 4-Dichlorobenzene	<10	<10	<10	<10	<10	<10				<10	P9/1	TM16/DM0
	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/DM0
	>10	>10	>10	10	>10	>10				>10	μg/ 1	TMAC/PM9
	<10	<10	<10	<10	<10	<10				<10	μg/ I	TN/16/PM9
2,6-Dinitrotoluene	<10	<10	<10	<10	<10	<10				<10	µg/ I	1M16/PM9
3-Nitroaniline	<10	<10	<10	<10	<10	<10				<10	µg/ I	FM16/PM9
4-Bromophenylphenylether	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
4-Chloroaniline	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
4-Chlorophenylphenylether	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
4-Nitroaniline	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
Azobenzene	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM9
Bis(2-chloroethoxy)methane	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
Bis(2-chloroethyl)ether	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/PM9
Carbazole	<10	<10	<10	<10	<10	<10				<10	µq/ I	TM16/PM9
Dibenzofuran	<10	<10	<10	<10	<10	<10				<10	ug/ I	TM16/PM9
Hexachlorobenzene	<10	<10	<10	<10	<10	<10				<10	µg/ I	TM16/PM0
Hexachlorobutadiene	<10	<10	<10	<10	<10	<10				<10	P9/1	TM16/DM0
	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM16/DM0
	>10	>10	>10	10	>10	>10				>10	μg/ 1	TMAC/PM9
nexactioroethane	<10	<10	<10	<10	<10	<10				<10	μg/ I	TM46/PM9
isophorone	<10	<10	<10	<10	<10	<10				<10	µg/ I	1W16/PM9
N-nitrosodi-n-propylamine	<10	<10	<10	<10	<10	<10				<10	µg/ I	ГМ16/PM9
Nitrobenzene	<10	<10	<10	<10	<10	<10		1	1	<10	ua/l	[ [M16/PM9
#### Jones Environmental Laboratory Client Name: O'Callaghan Moran & Associates

Client Name Reference: Location: Contact:

JE Job No.:

O'Callaghan Moran & Ass 09-188-01 Tipperary Town Landfill Barry Sexton 09/3874

#### **Report - Pesticides - waters**

 $\label{eq:Liquids/products: V=40ml vial, G=glass bottle, P=plastic bottle H=H_2SO_4, Z=ZnAc, N=NaOH, HN=HN0_3$ 

J E Sample No.	1-6	7-12	13-18	19-24	31-36	43-48						
Sample ID	MW1	MW2	MW3	MW4	MW6	MW8						
Depth	-	-	-	-	-	-						
COC No / misc										Please se	e attached i	notes for all
Containers	VHPG	VHPG	VHPG	VHPG	VHPG	VHPG				abbrev	iations and a	acronyms
Sample Date	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09	23/11/09						
Sample Type	Water	Water	Water	Water	Water	Water						
Batch Number	1	1	1	1	1	1						Method
Date of Receipt	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09	25/11/09				LOD	Units	No.
Combined Pesticide Suite												
Dichlorvos	< 0.01	< 0.01	<0.01	<0.01	< 0.01	< 0.01				<0.01	µq/l	TM042
Mevinphos	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				<0.01	ua/l	TM042
Alpha-BHC	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				<0.01	ua/l	TM042
Beta-BHC	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				< 0.01	ua/l	TM042
Gamma-BHC	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				< 0.01	ua/l	TM042
Diazinon	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				< 0.01	ua/l	TM042
Methyl Parathion	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01				< 0.01	ua/l	TM042
Ethyl Parathion (Parathion)	<0.01	< 0.01	<0.01	<0.01	< 0.01	<0.01				< 0.01	ug/l	TM042
Heptachlor	<0.01	< 0.01	<0.01	<0.01	< 0.01	<0.01				<0.01	ug/l	TM042
Fenitrothion	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01				<0.01	ug/l	TM042
Aldrin	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01				<0.01	ug/l	TM042
Malathion	<0.01	< 0.01	<0.01	<0.01	< 0.01	<0.01				<0.01	ug/l	TM042
Heptachlor Epoxide	<0.01	< 0.01	<0.01	<0.01	< 0.01	<0.01				< 0.01	ug/l	TM042
Endosulfan I	<0.01	< 0.01	<0.01	<0.01	< 0.01	< 0.01				<0.01	ug/l	TM042
Dieldrin	<0.01	< 0.01	<0.01	<0.01	< 0.01	<0.01				< 0.01	ug/l	TM042
4. 4'-DDE	<0.01	< 0.01	<0.01	<0.01	< 0.01	< 0.01		15°.		<0.01	ug/l	TM042
Endosulfan II	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01		. et		<0.01	ug/l	TM042
4 4'-DDD	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01		ott		<0.01	ug/l	TM042
Ethion	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	ald and			< 0.01	µg/l	TM042
Endrin	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	<0.01	orot			< 0.01	ua/l	TM042
Endosulfan Sulphate	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	<0.00	60,			< 0.01	ua/l	TM042
4.4'-DDT	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	\$0.01 N	<b>A</b> -			< 0.01	ua/l	TM042
Methoxychlor	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01	\$ <0.01				< 0.01	ua/l	TM042
Azinphos Methyl	< 0.01	< 0.01	< 0.01	< 0.01	<0.01	0.01				< 0.01	ua/l	TM042
Disulfoton	<0.01	< 0.01	<0.01	<0.01	<0.9	< 0.01				<0.01	ug/l	TM042
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### NOTES TO ACCOMPANY ALL SCHEDULES AND REPORTS

### SOILS

Please note we are only MCERTS accredited for sand, loam and clay and any other matrix is outside our scope of accreditation.

Where an MCERTS report has been requested, you will be notified within 48 hours of any samples that have been identified as being outside our MCERTS scope. As validation has been performed on clay, sand and loam, only samples that are predominantly these matrices, or combinations of them will be within our MCERTS scope. Your final report will reflect this, with non-MCERTS results on separate pages.

It is assumed that you have taken representative samples on site and require analysis on a representative subsample. Stones will generally be included unless we are requested to remove them.

All samples will be discarded one month after the date of reporting, unless we are instructed to the contrary. If we are instructed to keep samples, a storage charge of £1 (1.5 Euros) per sample per month will be applied until we are asked to dispose of them.

If you have not already done so, please send us a purchase order if this is required by your company.

Where appropriate please make sure that our detection limits are suitable for your needs, if they are not, please notify us immediately.

All analysis is carried out on as received samples and reported on a dry weight basis unless stated otherwise. Results are not surrogate corrected.

Asbestos screens where requested will be undertaken by a UKAS accredited laboratory.

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#### WATERS

otheruse Please note we are not a Drinking Water Inspectorate (DWI) Approved Laboratory . It is important that detection limits are carefully considered when requesting water analysis.

UKAS accreditation applies to tap water, surface water and groundwater only, any other liquids are outside our scope of accreditation

As surface waters require different sample preparation to groundwaters the laboratory must be informed of the water type when submitting samples. All samples are treated as groundwaters and anlalysis performed on settled samples unless we are instructed otherwise.

### **DEVIATING SAMPLES**

Samples must be received in a condition appropriate to the requested analyses. All samples should be submitted to the laboratory in suitable containers with sufficient ice packs to sustain an appropriate temperature for the requested analysis. If this is not the case you will be informed and any analysis that may be compromised highlighted on your schedule/ report by the use of a symbol.

The use of any of the following symbols indicates that the sample was deviating and the test result may be unreliable:

- sample temperature on receipt considered inappropriate for analysis requested \$
- ۸ samples exceeding recomended holding times
- samples received in inappropriate containers (e.g. volatile samples not submitted in VOC jars/vials) &
- no sampling date given, unable to confirm if samples are with acceptable holding times

### ABBREVIATIONS and ACRONYMS USED

- # UKAS accredited
- M MCERTS accredited
- NAD No Asbestos Detected
- ND None Detected (usually refers to VOC and/SVOC TICs)
- SS Calibrated against a single substance
- \* analysis subcontracted to a Jones Environmental approved laboratory.
- W Results expressed on as received basis
- Failed AQC results should be considered as indicative only and are not accredited.
- ++ Result outside calibration range, may be possible to re-run with higher detection limits

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10 DAN	4 DAY	Other	AVERAG	E COOL BO	X TEMP.(if require	:(p		-aj	Ž	00 00 7	かり	J J	ye ip	70	e.g. "High PAHs expected".	
5 DAY	3 DAY	]	SAMPLE	RECIEPT C	ONDITION:			2	30	or bo		21	20 7)	<u>}</u>	NOTE: If an M	CERTS report is
	SAMF	PLE INFORMATION (note: S = S	oil, W=Wa	iter, P=Pr	oduct/oil)			2	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	1500 JP US	17. 17. 17.	2	1 c Y, a	σν C	required this must	be requested whe
Lab ID	Ø	Sample ID	Matrix	Date	Time	Depth in Metres	Preservati	mi	ins 105	71	sat xci	19	109	nd nd	samples are	scheduled.
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Barry Sexton O'Callaghan Moran & Associates Granary House Rutland Street Cork Co.Cork Certificate No.: Job Ref: Sample Ref No.: LSN Page No.: Date Received: Date Reported: 352046 09K05195 38/66411 1 of 5 23/11/2009 24/11/2009

### TEST REPORT

Water - MW-4 - 23/11/09 -Sample Description Date Testing Initiated: 23/11/2009 MICRO 11:20 Category: Sample Condition: Satisfactory Order No.: NA Supplier Code: Method any other use. Unit Test Result Comments Est. MTC121 MPN/100mls **Total Coliform Count-**1,986 Colilert Consent of copyrig **E.COLI Count - Colilert** 10 MPN/400mls **MTC121** 

All tests are carried out according to our INAB schedule of accreditation.

Comments, opinions and interpretations expressed herein are outside this current scope of INAB accreditation. Results apply only to samples tested, and as received at the Laboratory.

Signed for and on behalf of Exova (Ireland) Ltd.

Michelle Everard B.Sc (Biosciences) Snr.Tech Microbiology Division



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Barry Sexton O'Callaghan Moran & Associates Granary House Rutland Street Cork Co.Cork Certificate No.: Job Ref: Sample Ref No.: LSN Page No.: Date Received: Date Reported: 352046 09K05195 38/66412 2 of 5 23/11/2009 24/11/2009

### TEST REPORT

Water - MW-5 - 23/11/09 -Sample Description Date Testing Initiated: 23/11/2009 11:35 MICRO Category: Sample Condition: Satisfactory Order No.: NA Supplier Code: Method any other use. Unit Test Result Comments Est. MTC121 MPN/100mls **Total Coliform Count-**205 Colilert Consent of copyrig **E.COLI Count - Colilert** MPN/400mls **MTC121** 21

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Signed for and on behalf of Exova (Ireland) Ltd.

Michelle Everard B.Sc (Biosciences) Snr.Tech Microbiology Division



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Barry Sexton O'Callaghan Moran & Associates Granary House Rutland Street Cork Co.Cork Certificate No.: Job Ref: Sample Ref No.: LSN Page No.: Date Received: Date Reported: 352046 09K05195 38/66413 3 of 5 23/11/2009 24/11/2009

# TEST REPORT

Water - MW-6 - 23/11/09 -Sample Description Date Testing Initiated: 23/11/2009 MICRO 11:45 Category: Sample Condition: Satisfactory Order No.: NA Supplier Code: Method any other use. Unit Test Result Comments Est. MTC121 MPN/100mls **Total Coliform Count-**2,420 Colilert Consent of copyrig **E.COLI Count - Colilert** 10 MPN/400mls **MTC121** 

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Signed for and on behalf of Exova (Ireland) Ltd.

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Barry Sexton O'Callaghan Moran & Associates Granary House Rutland Street Cork Co.Cork Certificate No.: Job Ref: Sample Ref No.: LSN Page No.: Date Received: Date Reported: 352046 09K05195 38/66414 4 of 5 23/11/2009 24/11/2009

# TEST REPORT

Water - MW-7 - 23/11/09 -Sample Description Date Testing Initiated: 23/11/2009 12:15 MICRO Category: Sample Condition: Satisfactory Order No.: NA Supplier Code: Method any other use. Unit Test Result Comments Est. MTC121 MPN/100mls **Total Coliform Count-**248 Colilert Consent of copyrig **E.COLI Count - Colilert** MPN/400mls **MTC121** 26

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Barry Sexton O'Callaghan Moran & Associates Granary House Rutland Street Cork Co.Cork Certificate No.: Job Ref: Sample Ref No.: LSN Page No.: Date Received: Date Reported: 352046 09K05195 38/66415 5 of 5 23/11/2009 24/11/2009

# TEST REPORT

Water - MW-8 - 23/11/09 -Sample Description Date Testing Initiated: 23/11/2009 11:55 MICRO Category: Sample Condition: Satisfactory Order No.: NA Supplier Code: Method any other use. Unit Test Result Comments Est. MTC121 MPN/100mls **Total Coliform Count-**1,986 Colilert Consent of copyrig **E.COLI Count - Colilert** 50 MPN/400mls **MTC121** 

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Signed for and on behalf of Exova (Ireland) Ltd.

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